DYNAMIC SIMULATION OF AN EXTRACTIVE

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DYNAMIC SIMULATION OF AN EXTRACTIVE

DISTILLATION COLUMN

by

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ABSTRACT

A mathematical model, based on heat transfer, mass transfer and fluid dynamics principles, has been developed to represent the dynamics of an extractive distillation column. The column simulation subroutine was incorporated into the recent DYNamic System Simulator executive package (DYNSYS 2.1), created at McMaster University, which uses Gear's method of integration.

Simulation runs were performed with step and pulse disturbances in the feed composition, feed flowrate and reflux ratio variables. Calculated results were compared with available experimental data obtained from a laboratory-scale (2 inch diameter), fifteen-stage extractive distillation column which used distilled water as the solvent to separate a binary mixture of acetone-methanol. Fairly good consistency was found within the normal operating conditions of the column.

A preliminary simulation study of the dynamics of the column (overhead composition transient behaviour) subject to controlled-cycling of the feed flowrate showed a substantial improvement in the separation efficiency of the column.

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CHAPTER 1

INTRODUCTION

Because of economic pressure, any industrial company is forced towards greater efficiency and productivity. In many areas, processes can be improved resulting in a favourable return on investment. One area for potential improvement is in process control.

Advances have been slow in industry because of the lack of mathematical techniques to analyze complex control systems and also because of the costly, sometimes impossible, implementation of advanced control schemes.

With the introduction of digital computers into the chemical process industries, more efficient control strategies have been investigated by major chemical companies. However, the approach, if it is to be widely used, has to be simple and fundamental in nature for practical application.

1.1 Computer control of distillation columns

Distillation has been one of the particular processes which attracted the many industrial control designers as well as university research control groups. Distillation can be simply understood as the process of vaporizing then condensing a mixture of components. Precise control of distillation is difficult to achieve because:

(1) Factors affecting separation are not readily interpreted in terms of control system requirements.

(2) Distillation processes are known to interact, sometimes severely.

(3) Measurements used for control may be non-existent or difficult to incorporate into the control strategy.

(4) Distillation units are usually the last in a chain of unit operations, therefore subject to upstream upsets.

(5) Towers with many trays respond very slowly, thus conventional negative feedback control does not seem to be the most efficient solution.

Poor control of a distillation column reduces the throughput capacity, makes the recovery of product difficult and results in excessive utility consumption.

The following sections will present a very practical approach to distillation control resulting from the study of Bojnowski (6).

1.1.1 Relative gain concept

Distillation is a multivariable, highly interactive process. From the control standpoint, a multivariable process is known to have a number of input variables called manipulators such as reflux rate, steam rate or bottom rate, and a number of output variables called controlled variables such as distillate product composition, column pressure or bottom composition.

In such a complex process, in order to achieve low operating cost, good operability and high product quality, the question should be "what variables do I want to control?". The answer is not so evident, the reason is that the controllers interact. Therefore, when looking at a multivariable process, often the question is asked: "If I change a manipulator, what

essential will change in the system?". Obviously, the problem is to determine the best pairing of controlled and manipulated variables. The expert control engineer can often choose the right pairs by experience, but sometîmes intuitive solutions can lead to dangerous consequences.

The technique of relative gain is commonly used to find the best pairing of variables. The approach bases the selection on a steady-state analysis which is probably the only drawback, since results must be checked against dynamic considerations. As said by Bojnowski (6), for most systems, the results are consistent with dynamic considerations.

The relative gain is defined as:

$$\lambda_{ij} = \frac{\partial c_i / \partial m_j |_m}{\partial c_i / \partial m_j |_c} = \frac{open \ loop \ gain}{c \ loop \ gain}$$
(1.1)

Physical meaning for a system of 2 × 2 can be found on fig.1.1. A change is forced in m_j by keeping all other m's constant, then the change in c_i is measured. This is the open-loop gain. The same change is applied to m_j , but all the c's, except c_i , are kept constant by closed-loop control; the change in c_i is observed then the closed-loop gain results.

Therefore, for a multivariable process, the relative gain matrix can be obtained:

$$\begin{array}{c} m_{1} & m_{2} & m_{j} \\ c_{1} & \begin{bmatrix} \lambda_{11} & \lambda_{12} \\ \lambda_{21} & \lambda_{22} \\ c_{i} & & & \lambda_{ij} \end{bmatrix}$$

(1,2)

Fig. 1-1 PHYSICAL MEANING OF THE RELATIVE GAIN IN THE MULTIVARIATE PROCESS



$$\frac{\partial c_1}{\partial m_1} \bigg|_{m_2} = \frac{\Delta c_1}{\Delta m_1} \bigg|_{m_2} = \text{open loop gas}$$

in





$$\frac{\partial c_1}{\partial m_1} \bigg|_{c_2} = \frac{\Delta c_1}{\Delta m_1} \bigg|_{c_2} = \text{closed loop gain}$$

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One property of the matrix is that the algebraic sum of any row or column is one. One must note that when λ_{ij} is one, m_j controls c_i and no interaction exists.

One does not have to perform both open-loop and closed-loop experiments. Obtaining the open-loop gains is sufficient since matrix manipulation will give the whole relative gain matrix. Techniques and examples are described in reference 6.

Applications can be based on the following rules:

(1) The pairs with the largest positive numbers should be selected for closed-loop control.

(2) Relative gain values approaching 0.5 indicate strong mutual coupling and will be extremely difficult to control.

(3) Negative numbers indicate a divergent control system, and if these pairs are chosen, control is impossible.

(4) One operating condition for which the c-m pairs were chosen may not be valid for another operating condition.

1.1.2 Eactors affecting distillation performance

The control strategy should keep a column functioning as designed, despite various upsets. Two very important facts must be retained for control design objectives:

(1) the composition of the distillate and bottoms streams are intimately affected by the product flowrates. If 100 lbs. of a component is fed to a column, 100 lbs. must be withdrawn. If the 100 lbs. is not withdrawn in the correct amount in each product stream, the desired composition cannot be obtained even though the column is capable of making the separation. This is called <u>the material balance problem</u>.

(2) Energy is required for separation. If there is not enough energy available to carry out a separation, even though the material flows are correct, the desired compositions will not be obtained. This is known as <u>the separation problem</u>.

Usually, control of the material balance is done with flow variables, and control of separation is achieved with reflux rate variable or feed point location. Unfortunately, in most cases, these two control loops severely interact. The problem is then to decouple the interaction by choosing the right pairs of controlled and manipulated variables.

Applying the relative gain technique and checking the performance on many industrial columns, Bojnowski mentioned many general rules for control of distillation:

(1) When only one product quality is to be controlled closely, the material balance must be manipulated.

(2) The lesser of two product streams should be manipulated to control product quality.

(3) The material balance should be manipulated to control the quality of the less pure product, and the separation should be manipulated to control the quality of the more pure product.

(4) The pressure must be tightly controlled for two reasons. If the pressure is increased, the liquid on the trays and in the reboiler is below its boiling point; the vapor is condensed to heat the liquid to its

boiling point; once at its boiling point, the vapor rate snaps back causing an increase in pressure. This causes column instability. Furthermore, if the column pressure is not held constant, measurements for control purposes are invalid. Suggested control schemes are described in reference 6.

1.1.3 Special features for extractive distillation control

In extractive distillation, the solvent is fed near the top of the column and entrains one or more heavy components on the way down the column. Light components concentrate up and are removed near the top. Heavy components and the solvent are withdrawn at the base and sent to a recovery column to recover the solvent which is recycled back to the extractive column.

When dealing with extractive distillation, one must take into account the following facts:

(1) Since the solvent is an extractant, the correct ratio between feed and solvent must be maintained. Insufficient solvent will cause poor separation while excessive solvent will result in the dilution of light components.

(2) Reflux greatly influences the separation. With higher reflux, better separation occurs, but excessive reflux dilutes the solvent thus causing poor separation. Thus a designer must try to control the solvent concentration at the solvent addition tray.

(3) Usually, extractive distillation is most sensitive to the solvent composition and temperature. Since the heat content of the solvent is a large percentage of total heat requirements, control of solvent temperature

is critical.

1.1.4 Conclusion

In summary, many methods of control other than feedback control have been used in distillation processes with certain special features. One example which can be mentioned is that of ultra high reflux and high reflux ratio separations; because of the extremely long response times, bang-bang control is the recommended control strategy since it is inexpensive, it gives extremely tight control, it gives fast responses to disturbances, it eliminates non-linearity problems such as sticking control valves and it eliminates tuning problems.

Feedforward control is usually applicable where:

(1) one column is noticeably limiting the capacity of the entire unit.

(2) a small improvement in recovery efficiency on a large refining column can mean substantial additional production.

(3) feed rate and composition changes upset a column early in a distillation train and cause problems in the rest of the operation.

(4) safety considerations demand that compositions stay within certain limits.

A fairly detailed presentation of different control schemes along with discussions can be found in reference 6.

1.2 Project objectives

In general terms, the present project is actually based on a fifteen stage column with thirteen trays and condenser and reboiler systems. Two feed streams enter the column: water, used as an extractant, enters near the top of the column, and a mixture of acetone-methanol is fed near the bottom. Simple binary distillation to separate acetone from methanol is not possible since the mixture forms an azeotrope of .8 mole fraction acetone at atmospheric pressure. As mentioned by D.O'Neill (10), at approximately 0.3 mole fraction water, the azeotrope is completely broken.

1.2.1 Long term objective on the equipment

A complete extractive distillation process must include two columns in series. The first one, usually called the extractive column, extracts the light component, in this case acetone. The second one, often called solvent recovery column, recovers the solvent which is recycled back to the extractive column.

The long term objectives will consist of building the second column, and studying some advanced control configurations on the complete process and checking the performance of each control scheme.

However, before doing any investigation of this kind on such a complex process, one must do more specific work to obtain the minimum requirements, e.g. to find out suitable models at different levels of sophistication, to achieve some simple control of input stream variables such as temperature or flowrate, and to understand the dynamic behavior of the process.

1.2.2 Summary of previous work

Jackson (11) undertook the initial step of this project. His work consisted of building the fifteen stage extractive column, evaluating its steady-state performance, interfacing basic column instrumentation to the mini-computer and developing a basic real time software package mainly for data logging and input stream temperature control. The software was made using the original version of the Real Time Operating System. Major operating problems occurred, due to tray priming at the solvent recovery section and tray weeping near to the bottom of the column.

D.O'Neill (10) rebuilt the column completely and evaluated its operating range as well as its steady-state performance. A steady-state simulation was also developed to better understand the process. A preliminary study on the dynamic behavior was made by disturbing input variables, such as feed composition and feed rate, by conventional step change and pulse change. Transient responses of overhead and bottom compositions were analyzed and satisfactorily fitted by first order transfer models with dead time.

1.2.3 Present objectives

The agreed long term objective was to investigate different control strategies to obtain the best separation performance. Digital computers become more and more an important tool for process designers. Industrial practice tends to use digital simulation to estimate roughly the performance of operations, rather than carrying out experiments on pilot plants. This, results in substantial time and research cost savings. Therefore, before investigating experimentally a control strategy, it is always economic to obtain a rough result by simulating this strategy on a computer. Obviously, the next step will be to develop a suitable mathematical model which can dynamically represent the extractive column since one needs process models to do any simulation work.

As far as the dynamic study is concerned, input disturbances must be achieved manually. With this technique, the study of oscillatory disturbances or random disturbances is difficult. Therefore, one of the main experimental steps will be to connect control valves into the two feedstreams. Consequently, input flowrates can be controlled automatically by the computer or can be disturbed, as programmed, by the computer.

A summary of the present objectives is:

(1) To develop a dynamic model representing the extractive distillation column

(2) To compare calculated step and pulse responses with available experimental data

(3) To perform a rough study with oscillatory disturbances, given the flexibility offered by the installation of the control values.

CHAPTER 2

PROCESS EQUIPMENT AND FACILITIES

As briefly mentioned in chapter 1, section 1.2, the column was totally rebuilt by D.O'Neill with the aim of eliminating liquid priming in the solvent recovery section, due to tray weeping and also due to technical problems caused by the vibrostaltic feed pumps during the operation.

The present chapter will give a brief description of the process equipment followed by the specific modifications made to improve the flexibility of the column for the dynamic study. Control software alterations will also be described as well as the specific possibilities of the modified process arrangement.

2.1 Previous apparatus

The complete details of the apparatus could be found in reference 10 chapter 4. A general description of the experimental equipment will be presented in this section.

2.1.1 Description

Fig. 2.1 shows the arrangement of the extractive distillation process (10) before the modifications.

Two storage tanks of 10 liters each, made of glass and located on the second floor of the chemical engineering laboratory, were used to supply

the feed, at two different proportions of acetone-methanol, to the column through 0.63 cm 0.D. stainless steel tubing. A system of on/off valves (V1, V2, V3, V4) allowed the sampling of the feed stream and the switching from one storage tank to the other. The tanks were in contact with the ambient atmosphere, therefore the feed just flowed down by gravity. A needle valve (V5) was used to adjust the flowrate manually. A filter prevented the penetration of foreign material to the glass orifice which was used in conjunction with a Dynascience Corporation differential pressure cell to measure the flowrate; the small signal from the DP cell was sent to a Dynascience Corporation Carrier-Demodulator which converted it into a 0-10 volts DC signal used for control purposes. A F α P precision flowrator indicated the current flowrate. Before entering the column, the feed was electrically heated by a preheater made from a 30.5 cm × 1.27 cm 0.D. tube wound with resistance wire. A thermocouple measured the outlet temperature from the preheater for control purposes, and a powerstat transformer could be used to vary the power input to the preheater. The feed line was connected to tray 11 from the top.

The same arrangement was used for the solvent stream, except that only one 10 liter storage tank was present. The tubing from the tank to the preheater input was in Tygon.

The column was made of thirteen sieve plates, a condenser system and a reboiler system. The sieve plates were separated by 15.25 cm × 5.08 cm I.D. Q.V.F. glass pipes. The weir limited the holdup of liquid on a tray; the overflow flowed down the downcomer to the tray below. Each sieve plate

fig. 2.1 FLOW DIAGRAM -EXTRACTIVE DISTILLATION COLUMN



was 0.16 cm thick and had 93 holes of 0.89 cm diameter; a feed port, vapor sampler, a thermocouple port and a liquid sampler were provided. A hot jacket surrounding the column to prevent heat loss to the ambient atmosphere was made from two sections of 13.34 cm I.D. perspex plastic pipe. A hot air gun blew the heated air upward in the jacket.

A stainless steel reboiler of 327 ml capacity was used with an overflow arrangement as shown in fig. 2.1; the liquid flowed by gravity or column pressure through the ice bath and went out by the tube exit which was maintained at a certain height that one can adjust to obtain the desired reboiler liquid level. The holdup of liquid from the output of the reboiler to the tube exit was 25 ml.

The condenser was equipped with a water-cooled system and a reflux divider electrically activated by a timing device made by Eagle Signal Corporation.

A more detailed description could be found in the work of Jackson (11) and of O'Neill (10).

2.1.2 Problems in operation and restrictions on dynamic study

The above equipment presented some problems encountered at different stages of the operation.

(1) Before doing any dynamic investigation on the process, one must bring the column to a steady-state level. From that point, perturbations can be put into the desired input variables. A practical question would be "when is the column at steady-state?" Until the present investigation, temperature readings at the stages and overhead flowrate measurements were used to determine experimentally whether the column is at steady state or not. Very often, when the samples were analyzed later on, it was found that a trend still existed as far as the composition is concerned, showing lack of steady-state reaching. If one wants to study the dynamic of the column due to a certain kind of input disturbance, this fact becomes crucial, since the response depends intimately upon the trend existing at the initial time of perturbation.

(2) It was noticed that during a run, it could happen that the air gun which blows heated air through the jacket, stopped for a significant period of time. This will undoubtedly cause an upset in the negligible heat loss assumption made in the simulation work.

Since each run usually requires at least 10 hours, the presence of electrical devices is very undesirable in the sense that the whole time for the experiment could be wasted if an electrical problem occurs at the last period of the planned run. The feed stream preheater, solvent stream preheater and reboiler heater are locations where electrical devices have been used.

(3) The needle values on the feed and solvent streams were used to adjust the flowrates to the desired values. When a flowrate, as measured by the DP cell, exceeded the tolerance limits, the computer sent a signal to the process site and the adjustment had to be carried out manually by checking the readings on the flowrators.

This technique of manual control is very inaccurate. From the dynamic point of view, since the operating range for the flowrates are small,

the possible perturbation amplitude is relatively small, therefore, this technique of controlling or varying the flowrate would not be recommended.

(4) Furthermore, oscillatory disturbances like sine wave, square wave or random disturbance, cannot be used as long as the computer is not arranged to do such perturbation. This is due to the inaccuracy of the manual adjustment and lack of time available for it since one must take samples quite often during a dynamic run.

Given the problems pointed out in the last two parts, some modifications were made in order to assist the investigator in obtaining better results as well as to give more versatility to the experimental dynamic work. The two next sections will present modifications on the equipment and on the computer control software.

2.2 Modifications to apparatus

2.2.1 Feed and solvent system

The needle valves V5 on the feed line and V7 on the solvent line were replaced by two control valves, 1/4 in. stainless steel body, trim 0, air to open type with neoprene diaphragm, manufactured by Badger Meter Inc. Fisher electro-pneumatic transducers, type 546, were used to receive a 1-9 VDC input signal from the D/A converter and to convert it into a pressure from 3-15 psi on the control valve diaphragm. An air line at 25 psig supplied the input to the transducers.

One problem occurred when replacing the needle valves by the control valves. The liquid coming from the storage tanks located about 15

feet above, flowed down just by gravity, and it was found that there was not enough head to push the liquid through the control valve orifice at the required rate even if it was fully opened. Two solutions were possible:

(1) One could design an electrical pump, and without altering the storage system, introduce it before the control valve location.

(2) One could replace the glass storage tanks by pressurized tanks and apply a pressure to create additional head on the liquid.

After considering the difficulties encountered while using electrical devices and the economic point of view, the second alternative was chosen.

2.2.2 Storage.tank system

Three stainless steel storage tanks were purchased from AC Tanks Co. two for the feed line and one for the solvent line. They could handle a maximum pressure of 200 psi and had a maximum capacity of about 25 liters. A nitrogen cylinder was used to exert the necessary pressure. The hardware connections are presented on fig. 2.2.

Such an arrangement implies that the pressure should be the same in the three tanks. Calibration curves for the feed line and the solvent line are presented in Appendix A. A safety valve (SV) was used for safety purposes and for clearing the pressure at the end of each run.

However, some operational difficulties occurred:

(1) One has to set the gas cylinder value at the pressure where the calibration was obtained, before each run, which is very difficult with the accuracy given by the actual pressure gauge; therefore calibration must be checked at every run.

(2) Filling the tanks requires a very substantial amount of time and caution since the mixture acetone-methanol is inflammable and usually a large amount of feed material is used. The feeding should be made from the top hole of the tanks. One must be extremely careful in estimating the quantity of material needed for the run since the tanks can be fed only before the run and not during the run.

The next section will present the modifications made to the computer program developed to assist the experimenter when operating the column.

2.3 Software facilities

2.3.1 Previous control package

D. Jackson (11) has developed a computer program to aid in the manual operation of the column. The software was created within a Real Time Operating System (RTOS) provided by the computer manufacturer. Computer-process interfacing allowed the function of input-output hardware devices according to a criterion which minimized the average time devices must wait for servicing.

Many tasks were created to initiate or terminate the operation, to perform parameter changes, to measure and display process temperatures, and input stream flowrates, to control the input stream temperatures according to an on/off scheme and to create a signal at the process site for manual correction of the input stream flowrates to desired values. The tasks were scheduled for execution by the operating system depending on their respective priority and the availability of the peripheral devices for input-output tasks.

Complete details of the software package as well as listings of programs and algorithms can be found in the work of D. Jackson (11). Since the software has to be entirely modified, no further description of the previous package is given in this work.

2.3.2 Modifications and new facilities

The actual computer system in the Chemical Engineering department is operating under the Real time Disk Operating System (RDOS) supplied by the computer manufacturer. References 23, 24, 25 are useful to understand how to use commands. All tasks developed to assist the operation of the column have been changed to be used with the Generalized Operating System Executive created by P. Tremblay (22). This executive offers a large number of facilities as well as a lot of flexibility for control purpose programs. Reference (22) must be regarded as the most important source of information on the actual control package for the distillation column.

Several programs were created:

(1) The program INFCE interfaces the user program which consists of a number of independent tasks, to the GOSEX.

(2) Tasks FEETC and SOLTC perform the control of the input feed and solvent temperatures by turning on or off the preheaters, and also

measure up to five temperature locations, convert into engineering units for displaying. There are many possible modes of display. The user should refer to reference (22) to be able to issue correct commands for these display modes.

(3) The introduction of the two control values in the feed and solvent lines required additional software to actuate the values via the computer. Two tasks were created, FEEFM and SOLFM, to calculate the control action to the value based on the flowrate reading from the D.P. cell and based on a PID control algorithm performed by the GOSEX utility routine CONTL. Instructions for using this routine could be found in detail in ref. 22. FEEFM and SOL-FM also convert the raw data, i.e. the stream flowrate, and the control action which is the percentage-opening of the value, into engineering units for display. The simple control loop for the two flowrates on the feed and solvent lines is presented on fig. 2.3.

(4) Task SSAMP was used to acknowledge at the computer site any change occurring at the process site by using the contact senses, for instance the taking of a sample by the operator.

(5) A program to input a square wave to the feed rate variable was also developed. Task FEEFT shifts the feed flowrate set point from a high level (FHL) to a low level (FLL) according to a half period determined by TWAIT. These parameters could be changed on-line by using the GOSEX function PARAM. (see ref. 22)

Algorithms and listings of the programs as well as the loading procedure are presented in Appendix G.





fig. 2.3 Elementary Flowrate Control Loop

2.4 Capabilities of modified apparatus and software

With the present apparatus, the study of the dynamics of the extractive distillation column could be extended to the introduction of oscillatory disturbances such as square wave and sine wave in the feed flowrate variable. The solvent flowrate variable could be investigated in the same way. A good variable to be studied will be the ratio of the feed to the solvent. Random disturbances, whether at two levels or according to a statistical distribution, could also be input into the input stream flowrates. With some modifications on the software, these disturbances could be realized under open-loop or under closed-loop control.

CHAPTER 3 DYNAMIC SIMULATION

As discussed in chapter 1, simulation on a digital computer has become and has been proven to be an efficient and economic tool in the design work of an engineer. Steady state simulation has been used intensively in university research as well as in industrial engineering work, especially to design new equipment, to conceive or improve processes, or to optimize the operating conditions for a better product quality at a minimum cost.

Dynamic simulation finds its importance, for example, in the study of optimal start-up of a process, in the investigation of the effects and the policy to be followed in the case of an equipment failure, and recently, especially in the transient study of process equipment and the design of different control strategies.

Before doing any simulation work, the first step must be to represent the real chemical equipment or process by an adequate mathematical relation. The cost of such model development will obviously increase with the degree of accuracy of the mathematical model; therefore such investment must be related to the objective of the simulation work.

3.1 Literature review on dynamic simulation of distillation

The literature contains many references to the study of the transient behavior of the distillation column which represents a complex and non-linear

chemical process.

Marshall and Pigford (7) presented a tentative modelling approach for packed columns. The continuity equation was applied to a differential unit length of the bed to yield differential material balance equations. A linear equilibrium relationship, constant molal holdup per unit length and constant molal overflow were the main assumptions.

Montroll and Newell (8) investigated the exact solution of a set of non-linear differential material balance equations for multistage separation of binary mixtures. On an IBM 701, Jackson and Pigford (9) tried to obtain a numerical solution for the differential material equations by approximating with finite difference equations. Total reflux, a relative volatility equilibrium relationship, a constant Murphree efficiency and a zero condenser hold-up were assumed. Also, the initial liquid compositions on the trays were considered as equal.

Huckaba and Danly (4) have done intensive studies on binary batch distillation which is basically a time dependent process. Besides the differential mass balance equations normally used by previous investigators, differential energy equations were introduced for the first time to take into account the inequality of the liquid flowrates between plates. The relaxation of the equality of flow assumption obviously contributed to a better representation of the real process.

Rosenbrock (5) developed mathematically a method of calculation of the unsteady state behavior of continuous distillation columns. The results for a step change in the feed composition were obtained for binary mixtures, based on total reflux assumption. Several non-linear equilibrium relationships

and a Murphree efficiency different from 100% were considered. Rosenbrock's calculated results were compared to the experimental data of Armstrong and Wilkinson (1), and it was found that the method was only acceptable for the initial transient period; fairly substantial divergences in the comparison appeared at later times. However, the use of a digital computer to handle complex and time-consuming calculation was found to be very promising.

An analytical exact solution to a step change in the feed composition was attempted by Voetter (12) using a linear equilibrium relationship. The introduction of the various simpligying assumptions was shown to yield up to 40 percent error in the results.

Acrivos (2) and Katz (3) have extended the dynamic study by introducing randomly fluctuating input conditions in a linear stagewise system with binary mixtures.

Franke (13) has modeled in detail the behaviour of a continuous experimental distillation column using a binary mixture, with a step change in the input feed composition.

The simulation of distillation with more than two components requires a much more powerful method to handle a large number of equations within a reasonable computer time. A typical problem could involve more than 50 components and columns with more than 100 stages.

A new technique of handling differential equations was used in the present study, with a tenary system and a fifteen stage column. Also, a new approach of using the energy balance equations to adjust the liquid and

vapor flowrates was investigated.

3.2 General features of the executive package DYNSYS 2.1

In chemical engineering, stiff systems of ordinary differential equations describing the process occur widely in reaction kinetics and in multistage processes. Stiff systems are defined as systems where the time constants differ greatly in magnitude. To be able to study the transient behaviour of such a system, the integration time step size is limited by the smallest time constant. Therefore the prediction of the entire dynamic response of such a system requires a large computer time and cost if conventional numerical techniques are used. An interesting investigation of different existing techniques was presented by Barney (14).

In the past few years, several executive packages have been developed especially for chemical engineers. These are more problem-oriented and use variations of what is known as the modular approach.

In 1970, Brobow, Johnson and Ponton have developed at McMaster University an executive package known as DYNamic SYStems Simulator, using a third order Adams-Moulton-Shell routine to integrate ordinary differential equations. This method proved to be insufficient to handle stiff systems. The integration technique was revised by Barney in 1974 and finally Gear's integration algorithm in conjunction with Key's technique of solving linear algebraic equations was implemented into the new version called DYNSYS 2.1. An option for stiff systems with a tridiagonal Jacobian matrix is also included since this case generally occurs when one deals with counter current stagewise processes.

The principal features of the executive DYNSYS 2.1 used in this project will be outlined in the following sections.

3.2.1 Modularity

A chemical plant is usually made up of processing equipment such as reactors, columns, heat exchangers, etc., with control devices such as analyzers, control valves, etc., connected by material flows and control or measuring signals.

Each equipment item or device can be represented by a mathematical equation or set of equations which form a unit computation or computation module. Then a mathematical model of a plant can be seen as a network of modules in which information will flow in a manner analogous to the material flow or control signals in the real plant. Therefore, the dynamic information flow diagram is very similar to the process flow diagram.

This simulation approach, known as the modular approach, is very convenient and flexible when one wants to study the combinations or configurations of physical equipment or to design control strategies on a digital computer. It is not as efficient computationally as the equationoriented approach since it involves more information transfer, but it helps the user to visualize his process, and especially it presents a great flexibility and an easy adaptability to different control and operational policies.

As a library of equipment and control modules gradually becomes available, the programming effort for new plant studies is reduced. However, the routines should be made general enough for further use. Another feature is that real variables are used in the modular approach, rather than transformed variables. The modules may be quite non-linear in behavior.

A detailed presentation of the concept can be found in reference 14.

3.2.2 Gear's method of integration

One of the most important features in a dynamic simulation executive is the numerical technique of integration. DYNSYS 2.1 adopted the method developed by C.W. Gear in 1971 to solve ordinary differential equations.

The subroutine is called DIFSUB(30) and uses a variable-order, variable step, linear, predictor-corrector algorithm. To be more flexible in handling a variety of types of chemical processes, the subroutine offers two options, one for stiff systems and one for nonstiff systems.

T

3.2.2.1 Nonstiff option

When the whole process comprises non-stiff modules, this option must be chosen. Basically, the predictor step is described by the Adams-Bashforth explicit equation:

$$y_{n+1} = y_n + h \sum_{i=1}^{k} \beta_i \dot{y}_{n+1-i}$$
 (3.2.a)

The corrector step uses the Adams-Moulton equation:

$$y_{n+1} = y_n + h \sum_{i=0}^{k} \beta_i' \dot{y}_{n+1-i}$$
 (3.2.b)
The coefficients β_i and β'_i may be found in reference 15, and the order may vary from 1 to 7.

3.2.2.2 Stiff option

When the whole process comprises just stiff modules, or both stiff and non-stiff modules, this option must be choosen. Obviously, the degree of stiffness is a matter for the judgement of the investigator.

The following approximations are used: for the predictor pass:

$$y_{n+1} = h \beta_1 \dot{y}_n + \sum_{i=1}^{k} \alpha_i y_{n+1-i}$$
 (3.2.c)

for the corrector pass:

÷

$$y_{n+1} = h' \beta_0^* \dot{y}_{n+1} + \sum_{i=1}^k \alpha_i^* y_{n+1-i}$$
 (3.2.d)

The coefficients in the predictor equation can be found in reference 15. Values of the corrector coefficients are given in reference 16. The order may vary from 1 to 6. The corrector equation has been found stable for stiff equations for order up to 6. Further details concerning accuracy, stability and stiffness coefficient determination are discussed in reference 14.

Since the corrector equation is implicit, a method of iteration is required for convergence of the solutions. Two methods of iteration are available for the stiff option: the direct Jacobi iteration is used for non-stiff modules in a stiff process, and the Newton-Raphson iteration is used for stiff modules. In the latter case, the Jacobian matrix is needed and may be user-supplied or evaluated numerically. Descriptions of the two iterative techniques can be found in reference 14. The application to the distillation column simulation is presented in section 3.3.

3.2.3 Algebraic equation solver

Since the corrector equation is implicit, a set of algebraiec linear equations will be obtained. An important factor in saving computer time cost is the choice of the method to solve these simultaneous equations, since one deals generally with large systems of ordinary differential equations in a plant simulation.

The coefficient matrix of the linear equations was shown to be a function of the Jacobian matrix for the system. In general, in a countercurrent stagewise process such as the distillation column, the Jacobian matrix has many zeros. Such a matrix is called a sparse matrix, and using a technique which just operates on the non-zero elements will reduce greatly the computer time used.

The routine called SIMULT developed by Key (28) in 1973 was adapted and implemented in DYNSYS 2.1. Seven pivoting options are available, but, the option of minimum row-minimum column was chosen for this project: the coefficient matrix is searched to find the row with the least number of nonzero coefficients; then for the non-zero elements in this row, the column with the most zeros is chosen as the pivotal column. If more than one row or column are chosen, the row or column with the smallest index is used.

The storage scheme is illustrated by the following example: let the matrix:

$$\underline{P} = \begin{bmatrix} 0 & 1 & 2 & 0 \\ 0 & 0 & 1 & 1 \\ 0 & 0 & 0 & 1 \\ 1 & 0 & 0 & 3 \end{bmatrix}$$

P is stored as:

$$\underline{A} = \begin{bmatrix} 1 & 2 \\ 1 & 1 \\ 0 & 1 \\ 1 & 3 \end{bmatrix} \qquad \underline{ICOL} = \begin{bmatrix} 2 & 3 \\ 3 & 4 \\ 0 & 5 \\ 1 & 4 \end{bmatrix}$$

One must refer to reference 14 for more details and explanations.

3.2.4 Adaptation of a new module to DYNSYS 2.1

This section will present the main features of how to construct a subroutine representing a unit computation and to adapt it to the main executive. This will be useful for future work when the simulation of the solvent recovery column must be made or when investigation on different control schemes is undertaken.

3.2.4.1 Main components of the DYNSYS 2.1 executive

Basically, the executive consists of the following programs:

- -DYN 1: this subroutine reads in data such as initial values of the variables, parameters of the equipment, characteristic values, etc., and prints out for checking, when called by the main program.
- -DYN 2: this subroutine establishes the logical sequence of execution of the modules, when called by the main program.
- -OUTPUT: this subroutine handles all outputs at each time interval specified by the user, when called by DYN 1 or by the main program.
- -PROPS: . this function reads in the properties of each component in the material flow when called by DYN 1.
- -DIFSUB: this subroutine uses the Gear's method to perform the integration and is called by the modules.
- -SIMULT: this subroutine solves the set of linear algebraic equations, simultaneously, when called by DIFSUB.
- -TRIDAG: this subroutine solves the set of linear algebraic equations simultaneously, when called by DIFSUB, if the Jacobian matrix is tridiagonal.

-DYNSYS: the main program of the executive.

The listings of the above subroutines can be found in reference 14. Fig. 3.1 shows the interactions between the different components involved in the simulation work in this project.



fig. 3.1 INTERACTION BETWEEN DYNSYS COMPONENTS

AND USER COMPONENTS

3.2.4.2 Construction of a module

Both the stiff and non-stiff options of Gear's method are available. However, the same option must be used for the entire simulation.

When the simulation diagram contains both stiff and non-stiff modules, the stiff option must be used, which means that the stiff coefficients in the approximative equations will be used in the integration. Then, depending on whether the module is stiff or non-stiff, the Newton-Raphson iteration or Jacobi direct iteration will be chosen.

A detailed skeleton of a module is shown in fig. 3.2.

Up to three iterations are allowed in the corrector pass. If convergence within a user-specified tolerance does not appear after the third iteration, the integration step size is reduced to 1/4 of its current size and the step is repeated for all modules.

The user can access to a number of parameters within the executive. A list is given in Appendix B.

3.3 Distillation column dynamic simulation

A tentative mathematical model has been written to represent the dynamic behavior of the experimental extractive distillation column of fifteen stages with a ternary organic system. The module created is used under the executive DYNSYS 2.1 for further flexibility for future work. The control configuration could be investigated after an appropriate model has been found for the main process.

fig. 3.2 SKELETON OF MODULE FOR DYNSYS 2.1

SUBROUTINE TYPE2.1

COMMENTS DESCRIBING MODULE

THE FOLLOWING COMMON BLOCKS ARE OR MAY BE REQUIRED: MAT, CON, PTAB, UNIT, GEAR2, MODULE, ROW, COLUMN, JACOB, SUBDI, DIAG, SUPERD

COMMON/MAT/MP(•), EP(•), S(• •), EX() COMMON/CON/IG, NCOMP, NC5, H, NE, NS, NPR, NPOL, TMAX, IURDER, NGRAPH COMMON/PTAB/IGFLAG, PP(•) COMMON/UNIT/IM, NMP COMMON/GEAR2/EPS, TIME, KFLAG, JSTART, NBVMAX, ICONV, ISTIFF COMMON/MODULE/IDERY, ITER, ITRI, MC, IPIVOT

THE ROW DIMENSION OF JCOL AND XJACOB MUST BE THE SAME AS IN DIFSUB MC IS MAXIMUM NUMBER OF COLUMNS IN XJACOB AND JCOL

COMMON/COLUMN/JCOL(,MC) (NORMAL OPTION) COMMON/JACOB/XJACOB(,MC) (NORMAL OPTION)

N IS NUMBER OF ODES

COMMON/SUBDI/A(N)	(TRIDIAGONAL	OPTION)
COMMON/DIAG/B(N)	(TRIDIAGONAL	OPTION)
COMMON/SUPERD/C(N)	(TRIDIAGONAL	OPTION)

CALCULATE MODULE PARAMETERS : STREAM INPUT, INITIAL CONDITIONS, VALUES OF ITER ETC.

INPUT STREAM INFORMATION IS OBTAINED FROM S(IG. .) VALUE OF INDEPENDENT VARIABLE (Y) NEED ONLY BE SPECIFIED ON FIRST INTEGRATION STEP (I.E. INITIAL CONDITIONS) JSTART=0 ON FIRST INTEGRATION STEP =CURRENT ORDER OF INTEGRATION TECHNIQUE ON LATER STEPS

IF ISTIFF=0,NONSTIFF COEFFICIENTS WILL BE USED IN INTEGRATION ALGURITHM (DIFSUB) FOR ALL MODULES, JACOBIAN MATRIX IS NOT REQUIRED

IF ISTIFF=1,STIFF COEFFICIENTS WILL BE USED FOR ALL MODULES THEN,IF ITER=0 DIRECT ITERATION OF CORRECTOR WILL BE USED (NONSTIFF MODULE) JACOBIAN IS NOT REQUIRED IF ITER=1 NEWTON-RAPHSON ITERATION OF CORRECTOR WILL BE USED (STIFF MODULE)

	JACOBIAN MATRIX WUST BE SUPPLIED
IC	
	ITER MUST BE SPECIFIED 0 OR 1 IT IS NOT USED UNLESS ISTIFF=1
¢C	IF(IG.EQ.2) GO TO 2
(C 4	00000000000000000000000000000000000000
+C	SECTION #2 : JACOBIAN EVALUATION
00	***************************************
3 3 3	SECTION #2 IS OMITTED FOR NONSTIFF MODULE
00	JACOBIAN MATRIX IS REQUIRED ONLY IF MODULE EQUATIONS ARE STIFF
¢C	CALCULATE JACOBIAN MATRIX ON CORRECTOR PASS ONLY
80	JACOBIAN NEED NOT BE VERY ACCURATE, AS IT IS USED ONLY FOR CONVERGENCE OF CORRECTOR
88	ONLY NONZERO ELEMENTS ARE CALCULATED,BUT ALL DIAGONAL ELEMENTS MUST RE STORED WHETHER OR NOT THEY ARE ZERO (NORMAL OPTION)
88	NORMAL OPTION
ĉ	IF JACOBIAN MATRIX IS NOT TRIDIAGONAL KEYS STORAGE SCHEME IS USED
ŝ	COLUMN NUMBERS OF NONZERO ELEMENTS IN ROW I ARE STORED IN
	SIMILARLY THE VALUES OF THE NONZERO ELEMENTS OF ROW I ARE
œ	STORED IN XJACOB(I,)
С С	REMEMBER TO STORE ALL DIAGONAL ELEMENTS EVEN IF THEY ARE ZERO
888888	MC - MAXIMUM NUMBER OF COLUMNS IN XJACOB MATRIX MC CAN BE LARGER THAN THE MAXIMUM NUMBER OF NONZERO ELEMENTS IN ANY JACOBIAN ROW SINCE ADDITIONAL NONZERO ELEMENTS CAN BE CREATED DURING THE SOLUTION.MC IS CONSTANT FOR ANY SET OF ODES AND MAY HAVE TO BE DETERMINED BY TRIAL AND ERROR
E E	IPIVOT - PIVOT OPTION USED IN SIMULT (1-7)
ĉ	IPIVOTEZ I GAUSS-JORDAN ELIMINATION IPIVOTEZ I GAUSS-JORDAN PARTIAL PIVOTING
ũ	IPIVOT=3 : GAUSS=JORDAN FULL PIVOTING
C	IPIVOT=4 : MINIMUM ROW-MINIMUM COLUMN
E C	IPIVOT=5 : MINIMUM COLUMN-MINIMUM ROW
č	IPIVUI=6 I MAXIMUM CULUMN=MINIMUM RUW IPIVUI=7 I MINIMUM OF ROW ENTRIES TIMES COLUMN ENTRIES
C	TITADI-1 - HININGH OF KOW ENHALES FIMED COLOUN ENHALES
888	ICOL(I.J) - COLUMN NUMBER OF JTH NONZERO ELEMENT IN ROW I XJACOB(I.J) - VALUE OF JTH NONZERO ELEMENT OF ROW I
-	MC=
	IPIVOT=
	IROW(1.1)=
	XJACOB(1+1) =
	•

IROW(N,)= XJACOB(N.) =

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TRIDIAGONAL OPTION

FOR TRIDIAGONAL JACOBIAN MATRIX, THE SUBDIAGONAL, DIAGONAL AND SUPERDIAGO ELEMENT VALUES ARE STORED IN ARRAYS ADB AND C FROM CUMMON BLOCKS SUBDI. DIAG AND SUPERD RESPECTIVELY

N - NUMBER OF ODES - VALUES OF SUBDIAGONAL ELEMENTS ARE STORED IN A(2) ... A(N) A(1) IS NOT USED - VALUES OF DIAGONAL ELEMENTS B - VALUES OF SUPERDIAGONAL ELEMENTS ARE STORED IN C(1) ... C(N=1) С C(N) IS NOT USED 8(1)= C(1) =A(2)= B(2)= C(2) =A(N-1) =B(N-1) =C(N-1) =A (N) = B(N)= 1 CONTINUE IF TRIDIAGONAL OPTION IS BEING USED.ITRI MUST BE SET TO 1 HERE. IF NORMAL OPTION IS USED ITRI MAY BE IGNORED ITRI=1 2 CONTINUE Cee SECTION #3 : DERIVATIVE CALCULATION CALCULATE DERIVATIVES DERY(1) =DERY(N)= $\mathbf{C}_{\mathbf{e}_{\mathbf{e}}_{\mathbf{$

С SECTION #4 : CALL DIFSUB С С CALL DIFSUB TO SOLVE ODES FOR MODULE 00000000 DIFSUB MAY BE CALLED MORE THAN ONCE FROM A MODULE FOR EXAMPLE IF THE MODILE CONTAINS A SET OF STIFF O.D.E.S AND ANOTHER SET OF NONSTIFF 0.D.E.S ARGUMENTS : - NUMBER OF ODES N Y - INDEPENDENT VARIABLE DERY - DERIVATIVES CALL DIFSUB (N.Y.DERY) CCCC IF IDERY IS NOT ZERO, THE DERIVATIVES WILL BE RE-EVALUATED AND RETURNED TO DIFSUB ITRI MUST ALSO BE RESET IF IT IS 1 C IF(IDERY NE.0) GO TO 1 C SECTION #5 : STREAM OUTPUT CALCULATION С CALCULATE STREAM OUTPUT (STORED IN S(1, ,)) C S(1. .)= RETURN

END

3.3.1 Assumptions

The continuous distillation of a liquid mixture in either a stagewise or a continuous-contact column is a very complex phenomenon which involves heat transfer, mass transfer and fluid dynamics. To obtain complete description of such a process, one must formulate the mechanics and interrelationships of these three transport phenomena.

The mechanics of the fluid flow requires the understanding of the column internals, the pipe sizes and arrangement, the condenser system and the reboiler system. The liquid and vapor hold-ups could be determined thereafter using some empirical relations. The next step would be to set the relationship between the vapor and the liquid at these mixing regions. Considering, the liquid holdup on a plate of the column, the composition is a function of time and position within the liquid phase on the plate. The mass transfer rate is basically a function of the difference in composition of the vapor and liquid phases at this point and also of the composition gradient within the liquid phase.

Consequently, it is apparent that a mathematical dynamic model obtained rigorously will be very complex and hard to use.

However, before modeling is attempted for a process, the objectives of such work must be established. The investment in modeling will therefore be determined by the probable return to be gained by using the model.

In the actual project, the main purposes of constructing a dynamic simulation for the distillation column are to predict the transient behavior of the column in response to a variety of types of input disturbances, to investigate the response when the column is disturbed by combinations of possible input variables and to design feasible control strategies as well as to obtain an acceptable design of the control parameters and systems dynamic parameters. If these objectives can be achieved satisfactorily from a relatively simple model, then it is apparent that the pursuit of a completely rigorous model is unwarranted. If the model proves to be insufficient for some of the above objectives, the model could be improved by deleting the least probable assumptions.

The following assumptions have been made in developing the mathematical model of the actual column's transient behavior:

(1) There is no heat loss along the column. This assumption could be shown valid to a certain extent since the column is surrounded by a heated air jacket to minimize the heat transfer from inside the column to outside.

(2) The fluid dynamic response time is negligible compared with that of mass transfer. This assumption is justified when no substantial changes in liquid flowrates are experienced, which is very appropriate for changes in feed composition. Also, the range of input flowrate variation is found to be quite small. Moreover, the work of May (17) indicated that for small towers, the time to reach fluid equilibrium is essentially negligible compared to that for mass transfer. The diameter of the present column is 2 inches.

(3) The molar vapor holdup is small compared with that of the liquid. Huckaba and Danly (4) have shown that this assumption is perfectly valid for atmospheric distillation.

(4) The volumetric liquid holdup at each tray is not a function of time. The initial holdups are estimated by the Francis Weir relationship, using the initial liquid flowrates as the basic information. The equations are given in Appendix C.

(5) The flowing fluids are all single phase streams. The liquid and vapor entrainment are negligible.

(6) The pressure through the column is constant and equals atmospheric pressure. This assumption is valid in this case if one considers the small number of plates and the relatively small height of the liquid layers. This assumption could be deleted to make the pressure a linear function with position along the column. However, no big improvement is expected.

(7) The plate efficiencies are independent of time. The work of O'Neill(10) indicated that the efficiencies could be kept constant within the actual operating range without affecting drastically the results of the steady state simulation.

(8) The volumetric reboiler and condenser hold-ups are estimated experimentally. A 100% mixing is assumed in both holdups. The assumption for the reboiler is questionable.

(9) The liquid composition at each tray is assumed uniform. This could be valid since the diameter of the tray and the weir height are reason-ably small.

Based on the above assumptions, the basic equations will be presented in the next section.

3.3.2 Basic equation derivation

3.3.2.1 Material balance equations

The basic principle of conservation has been applied to each tray including the feed tray and the solvent input tray and also to the reboiler and the condenser. All the equations have been derived using the gram-mole as the material unit. The starting point of the derivation is the generalized statement:

Input - output + generation - disappearance = accumulation Based on the nomenclature in fig. 3.3, one can write for a ternary system:

For a normal tray:

$$\frac{dM_n}{dt} = -V_n - L_n + L_{n-1} + V_{n+1}$$
(3.3.1.a)

$$\frac{d(M_{n}x_{n,i})}{dt} = -V_{n}y_{n,i} - L_{n}x_{n,i} + L_{n-1}x_{n-1,i} + V_{n+1}y_{n+1,i}$$
(3.3.1.b)

For a feed tray:

$$\frac{d}{dt}M_{n} = -V_{n} - L_{n} + L_{n-1} + V_{n+1} + F \qquad (3.3.2.a)$$

$$\frac{d}{dt} (M_{n}x_{n,1}) = V_{n}y_{n,1} - L_{n}x_{n,1} + L_{n-1}x_{n-1} + V_{n+1}y_{n+1,1} + Fx_{f,1}$$
(3.3.2.b)

For the condenser (n=1)

$$\frac{d}{dt} M_n = V_{n+1} - D - R \qquad (3.3.3.a)$$

$$\frac{d}{dt} (M_n x_{n,i}) = V_{n+1} y_{n+1,i} - Dx_{n,i} - Rx_{n,i}$$
(3.3.3.b)

i = 1, 2

For the reboiler (n = 15)

$$\frac{d}{dt}M_{n} = -V_{n} + L_{n-1} - B \qquad (3.3.4.a)$$

$$\frac{d}{dt} (M_n x_{n,i}) = -V_n y_{n,i} + L_{n-1} x_{n-1,i} = Bx_{n,i}$$
(3.3.4.b)

For convenience, the compositions $x_{n,i}$ were used as variables of integration. Therefore, the equations (3.3._.b) were transformed using the equations (3.3._.a) to obtain on the left hand-side the derivative of the composition $x_{n,i}$ as a function of time.



After simple manipulations, the following equations were obtained: For a normal tray:

$$\frac{d}{dt} x_{n,i} = \frac{-V_n y_{n,i} + L_{n-1} x_{n-1,i} + V_{n+1} y_{n+1,i} - M_n}{M_n}$$

$$\frac{x_{n,i} (-V_n + L_{n-1} + V_{n+1})}{M_n}$$

$$i = 1, 2 \quad (3.3.1.c)$$

For a feed tray:

$$\frac{d}{dt} x_{n,i} = \frac{-V_n y_{n,i} + L_{n-1} x_{n-1,i} + V_{n+1} y_{n+1,i} + M_n}{M_n}$$

$$\frac{F x_{f,i} - x_{n,i} (-V_n + L_{n-1} + V_{n+1} + F)}{M_n}$$

$$i = 1, 2 \quad (3.3.2.c)$$

For the condenser (n=1)

$$\frac{d}{dt} x_{n,i} = \frac{(y_{n+1,i} - x_{n,i}) V_{n+1}}{M_n} \qquad i = 1,2 \quad (3.3.3.c)$$

For the reboiler (n=15)

$$\frac{d}{dt} x_{n,i} = \frac{-V_n y_{n,i} + L_{n-1} x_{n-1,i} - x_{n,i} (-V_n + L_{n-1})}{M_n}$$

i = 1, 2 (3.3.4.c)

The 30 equations (3.3._c) were integrated as a function of time, and the third composition (i=3) was calculated according to:

In this simulation program, the integration was made on the acetone composition (i=1) and the methanol composition (i=2); the water composition (i=3) was obtained by difference.

3.3.2.2 Vapor-liquid equilibrium relations

The activity coefficient of each component in the liquid mixture was calculated using the enthalpic Wilson equation, extended to multicomponent mixtures.

Given a liquid mixture of q components, the activity coefficient of component i in the mixture was calculated from:

$$\ell n_{\gamma_{i}} = -\sum_{\substack{j=1\\j\neq i}}^{q} \frac{x_{j} \ell n (\Lambda_{ij}\Lambda_{ji})}{(\sum_{j=1}^{p} (\sum_{j=1}^{n} (\sum_{j$$

$$\sum_{\substack{m,j\\m\neq j\neq i}} \frac{x_j x_m \ell n(\Lambda_{jm} \Lambda_{mj})}{\sum_{\substack{m,j\\p \neq p}} x_p (\Sigma \Lambda_{mp} X_p)} \times \alpha_{ijm} \qquad (3.3.6.a)$$

where

$$\alpha_{ij} = 1 + x_i \left(1 - \frac{1}{\sum_{p} \Lambda_{ij} x_p} - \frac{\Lambda_{ji}}{\sum_{p} \Lambda_{jp} x_p} \right)$$
(3.3.6.b)

$$\alpha_{ijm} = 1 - \Lambda_{ji} / (\Sigma \Lambda_{jp} x_p) - \Lambda_{mi} / (\Sigma \Lambda_{mp} x_p)$$
(3.3.6.c)

$$\Lambda_{ii} = 1$$
 (3.3.6.d)

$$\Lambda_{ij} = \frac{v_j}{v_i} \times \exp(\frac{-\lambda_{ij} - \lambda_{ii}}{RT})$$
(3.3.6.e)

The parameters λ_{ij} for each pair of components for the present mixture of acetone-methanol-water could be found in Reference 20. Griswold and Buford (21) have tested the above equations for the ternary system mentioned above. The derivation of the enthalpic Wilson equations is presented in reference 20. The v_i 's are the molar volume of the components at temperature T in °K.

Once the activity coefficient of each component was calculated, the activity r_i could be derived by the following relation:

$$\Gamma_i = \frac{p_i}{p} \times \gamma_i$$
, p is the solution pressure (3.3.7)

where

 $p_i = \exp(a + \frac{b}{1+c}), p_i$ is the vapour pressure (3.3.8)

The composition of the vapor in equilibrium with the liquid mixture y*

was then calculated from:

$$y_{i}^{*} = \Gamma_{i} x_{i}$$
 (3.3.9)

The temperature of the liquid on each tray was used as the parameter to converge the summation of the vapor component compositions to 1.

The Murphree efficiencies $E_{n,i}$ could then be applied to calculate the true composition of the vapor leaving each tray. The following relation was used for tray n:

$$y_{n,i} = y_{n+1,i} + E_{n,i} \times (y_{n,i}^* - y_{n+1,i})$$
 (3.3.10.a)
n=1,14

As no vapor enters the reboiler (n=15), the Murphree efficiency was defined differently by taking the composition of the liquid leaving the reboiler instead of the composition of the vapor coming to the stage. Therefore, one obtains for the reboiler:

$$y_{n,i} = x_{n,i} + E_{n,i} \times (y_{n,i}^* - x_{n,i})$$
 (3.3.10.b)
n=15

3.3.2.3. Energy equations

At the end of each time interval, the new compositions of the liquid and the vapor leaving and reaching the stage were obtained as shown in the previous sections. Then, the unsteady-state energy equation at each stage was used to readjust the vapor and liquid flowrates.

The energy imbalance at tray n was obtained according to the following principle:

energy accumulation = energy in - energy out

$$\begin{pmatrix} dH_{n} \\ dt \end{pmatrix}_{t}^{=} (H_{L,n-1}L_{n-1} - H_{L,n}L_{n} + H_{V,n+1}V_{n+1} - H_{V,n}V_{n} + Q_{n})_{t}$$

$$(3.3.11)$$

This surplus of energy was assumed to vaporize a certain quantity of liquid from the tray holdup, thus increasing the flowrate of the vapor leaving this tray. The vapor material amount was obtained from:

$$\frac{(dV_n)}{dt}_t = \frac{(dH_n/dt)_t}{(H_{V,n} - H_{L,n})_t}$$
(3.3.12)

Therefore, the vapor flowrates could be readjusted, starting from the reboiler up the column.

At the top of the column, due to the change in the vapor flowrates, the unbalance of the liquid flow could be computed and flowed down, trying to keep the energy equation instantaneously balanced.

3.3.3. Jacobian matrix derivation

As mentioned previously, the integration procedure requires the

Jacobian matrix of the system of ordinary differential equations, to perform the convergence of the solutions at a certain time t, in the corrector pass.

In general, for the system y = f(y), the Jacobian matrix is:

$$\underline{J} = \begin{bmatrix} \frac{\partial f_1}{\partial y_1} & \frac{\partial f_1}{\partial y_2} & \frac{\partial f_1}{\partial y_n} \\ \frac{\partial f_2}{\partial y_1} & \frac{\partial f_2}{\partial y_2} & \frac{\partial f_2}{\partial y_1} \\ \vdots & \vdots \\ \frac{\partial f_n}{\partial y_1} & \frac{\partial f_n}{\partial y_n} \end{bmatrix}$$

Taking the partial derivative of the right hand side of equations (3.3.1.c) (3.3.2.c), (3.3.3.c) and (3.3.4.c) with respect to $x_{n-1,i}$ and $x_{n,i}$ successively, one could obtain for a normal tray:

$$\frac{d}{dx_{n-1,i}} \left(\frac{dx_{n,i}}{dt}\right) = \frac{L_{n-1}}{M_n} \qquad i = 1, 2 \quad (3.3.13.a)$$

$$\frac{d}{dx_{n,i}} \left(\frac{dx_{n,i}}{dt} \right) = - \frac{\left(-V_n + L_{n-1} + V_{n+1} \right)}{M_n}$$

i = 1, 2 (3.3.13.b)

for a feed tray:

$$\frac{d}{dx_{n-1,i}} \left(\frac{dx_{n,i}}{dt}\right) = \frac{L_{n-1}}{M_n} \qquad i = 1, 2 \quad (3.3.14.a)$$

$$\frac{d}{dx_{n,i}} \left(\frac{dx_{n,i}}{dt}\right) = \frac{-(-V_n + L_{n-1} + V_{n+1} + F)}{M_n}$$

$$i = 1, 2 \quad (3.3.14.b)$$

for the condenser (n=1)

$$\frac{d}{dx_{n-1,i}} \left(\frac{dx_{n,i}}{dt} \right) = 0 \qquad i = 1, 2 \quad (3.3.15.a)$$

$$\frac{d}{dx_{n,i}} \left(\frac{dx_{n,i}}{dt}\right) = \frac{-V_{n+1}}{M_n} \qquad i = 1, 2 \quad (3.3.15.b)$$

for the reboiler (n=15)

$$\frac{d}{dx_{n-1,i}} \left(\frac{dx_{n,i}}{dt}\right) = \frac{L_{n-1}}{M_n} \qquad i = 1, 2 \quad (3.3.16.a)$$

$$\frac{d}{dx_{n,i}} \left(\frac{dx_{n,i}}{dt}\right) = \frac{-(-V_n + L_{n-1})}{M_n} \qquad i = 1, 2 \quad (3.3.16.b)$$

It is apparent to find that the Jacobian matrix has a tridiagonal form.



3.3.4 Simulation software

A module representing dynamically the experimental distillation column has been created and incorporated to the DYNSYS 2.1 executive. A logic flowchart of the calculation procedure could be found on fig. 3.4.

A stand-alone steady-state program was developed by O'Neill (10) and was used to supply the starting steady-state conditions for the dynamic simulation. Information related to the utilization of the steadystate package was presented in detail in reference 10.

A data transformer program was created to adapt the output of the steady-state program to a suitable format to input into the dynamic simulation program.

The dynamic runs were performed on a CDC 6400 (McMaster University). Listings of the utility subroutines can be found in Appendix F. Listing of the main subroutine is presented in Appendix F.

CHAPTER 4

DYNAMIC RESULTS - STEP AND PULSE DISTURBANCES

The immediate objective of the work was to develop a mathematical model to represent adequately the experimental extractive distillation column. In order to evaluate the present simulation algorithm, conventional disturbances such as step changes and pulse changes, were introduced into input variables such as feed flowrate, feed composition and reflux ratio. The dynamic simulated response has been observed and compared to available experimental results obtained by D. O'Neill (10).

A presentation and analysis of the calculated and experimental results are given in the following sections, for disturbances in feed composition, feed flowrate and reflux ratio respectively.

4.1 Dynamic responses due to feed composition disturbances

In order to observe the dynamic response of the column, especially the transient behavior of the overhead product composition which generally is the controlled variable, the feed acetone composition was disturbed by a negative step change of -5.3% amplitude, a positive step change of +5.6% amplitude and a negative pulse change of -17.4% amplitude. The operating conditions of the three runs could be found on Table 4.1.

Experimentally, the column was brought initially to a certain steadystate level and the disturbance was introduced. A system of on/off valves

permits a switch of the feed line from one storage tank to another storage tank with different composition. Samples were taken from the overhead line and from the bottom line, more frequently at the beginning of the change, and were analyzed by a gas chromatograph.

Plots of experimental and simulated results with respect to time were made and presented on fig. 4.1, 4.2, 4.3. The variation of the temperature was found negligible in both experimental and simulated results. The comparison was found more consistent for the case of a pulse change. Bottom product compositions were apparently insensitive to feed composition changes as shown by experimentation and simulation.

4.2 Dynamic responses due to feed flowrate disturbances

The effects of a change in the feed flowrate were investigated. Such a change caused a perturbation in the overall mass balance of the column, therefore affecting the local and component mass balances throughout the stages.

Experimentally, a needle valve allows the manual introduction of a negative step change of -15.4% amplitude, a positive step change of +18.2% amplitude and a positive pulse change of +18.2% amplitude in the feed flow-rate. The operating conditions of the three runs are presented in Table 4.2.

In the simulation, the mass change of the feed was supposed to appear instantaneously at the bottom product line, for the total mass balance to be satisfied. Therefore, sudden changes occurred for all liquid flowrates from the feed tray downward to the reboiler. Thus, the transient of the

TABLE 4.1 OPERATING CONDITIONS

FEED COMPOSITION AS DISTURBED VARIABLE

Quantity	Unit	negative step change		positive step change		negative pulse change	
Quantity		initial	final	initial	final	initial	final
F	g moles/hr	16.16	16.16	16.16	16.16	16.16	16.16
× _{F,1}	mol. fract.	.6076	.5753	. 5753	.6076	.6067	.6076
× _{F,2}	mol. fract.	.3848	.4171	.4171	.3848	.3848	.3848
т _F	°C	45	45	45	45	45	45
S	g moles/hr	30.75	30.75	30.75	30.75	30.75	30.75
×s,1	mol. fract	. 0.	0.	0.	0.	0.	0.
× _{S,2}	mol. fract	. 0.	0.	0.	0.	0.	0.
T _S	°C	70	70	70	70	70	70
RR	l dimension- less	1.756	1.756	1.756	1.756	1.756	1.756
REBDT	Kcal/hr	235	235	235	235	235	235
%dist. AMPL	dimension- less	-5.3 %		+	5.6 %	- 1	7.4 %



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column resulted from the imbalance of the component mass equation for the trays between the feed and the reboiler. The fluid dynamic equilibrium time was assumed negligible compared to the equilibrium time of the temp-erature or composition.

The transient responses at the overhead and at the bottom were presented on fig. 4.4.4.5, 4.6. An increase in the feed flowrate means an increase in the quantity of acetone and methanol. Since most of the acetone goes up to the condenser, the acetone mole fraction at the overhead should increase relatively to the methanol mole fraction. However, increases of both acetone mole fraction and methanol mole fraction at the reboiler could be explained by the fact that there was more acetone than methanol in the feed line.

4.3 Dynamic responses due to reflux ratio changes

An increase in the reflux ratio at the overhead implies a decrease in the overhead flowrate and vice versa. The results were simulated for the cases of a negative step change of -11.3% amplitude, a positive step change of +12.7% amplitude and a positive pulse change of +75% amplitude.

Once again, the overall mass balance of the column was affected. In the simulation, when a change in the reflux ratio occurred, the total mass balances on each stage were completed by adjusting instantaneously the liquid flowrates leaving each stage. Therefore, an imbalance of the component mass equations resulted and implied a change in the composition of the liquid.

Experimentally, a reflux divider allowed the introduction of such

TABLE 4.2 OPERATING CONDITIONS

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FEED FLOWRATE AS DISTURBED VARIABLE

	unit	negative step change		positive step change		positive pulse change	
Quantity		initial	final	initial	final	initial	final
F	g moles/hr	17.53	14.83	14.83	17.53	14.83	14.83
× _{F,1}	mol. fract.	.6059	.6059	.6059	.6059	.6059	.6059
× _{F,2}	mol. fract.	.3875	.3875	.3875	.3875	.3875	.3875
т _F	°C	45	45	45	45	45	45
S	mol. fract.	30.75	30.75	30.75	30.75	30.75	30.75
× _{S,1}	mol. fract.	0.	0.	0.	0.	0.	0.
×s,2	mol. fract.	0.	0.	0.	0.	0.	0.
Τ _S	°C	70	70	70	70	70	70
RR	dimension-	1.784	1.784	1.784	1.784	1.784	1.784
REBDT	Kcal/hr	234	234	234	234	234	234
% dist.amp	dimension- less	- 15.8 %		+ 18.2 %		+ 18.2 %	



% MOLE FRACTION




TABLE 4.3 OPERATING CONDITIONS

.1

REFLUX RATIO AS DISTURBED VARIABLE

1

Quantity	unit	negative st	ep change	positive step change		positive pulse change	
		initial	final	initial	final	initial	final
F	g moles/hr	16.37	16.37	16.37	16.37	16.37	16.37
× _{F,1}	mol.fract.	.5886	.5886	.5886	.5886	.5886	.5886
× _{F,2}	mol.fract.	.4060	.4060	.4060	.4060	.4060	.4060
т _F	°C	45	45	45	45	45	45
S	g moles/hr	30.75	30.75	30.75	30.75	30.75	30.75
×s,1	mol. fract.	0.	0.	0.	0.	0.	0.
×s,2	mol. fract.	0.	0.	0.	0.	0.	0.
T _S	°C	70	70	70	70	70	70
RR	dimension- less	1.756	1.558	1.558	1.756	1.756	1.756
REBDT	Kcal/hr	215	215	215	215	215	215
%Dist.ampl	dimension- less	-	11.3 %	+ 12.	7 %	+ 75	ž



4 . . .









changes to the reflux ratio variable. The operating conditions are presented in Table 4.3.

The calculated results were compared to the available experimental data and were presented on fig. 4.7, 4.8, 4.9. A fairly good agreement was obtained at the overhead. However, inconsistencies were observed at the bottom of the column. These could be likely explained by the assumption of 100 per cent mixing the reboiler not being satisfied in practice.

4.4 Conclusion

The above study has shown that the dynamic behaviour of a fifteenstage extractive distillation column can be described by simply laying out the basic mass, energy and equilibrium equations and by choosing an appropriate set of assumptions. The Murphree efficiencies at the stages were kept constant throughout the simulation and came from D. O'Neill's steady-state simulation work (19). The parameters in the enthalpic Wilson equation extended to multicomponent mixtures were taken from Bruin's study (20) about activity coefficient relations in miscible and partially miscible multicomponent systems. Only the parameter in the Francis weir relation was adjusted to obtain a fair consistency in the comparison of simulated data with experimental data. This relation was used to calculate the overflow height as a function of the tray characteristics and the liquid flowrate. A change in the overflow height will cause the total tray holdup to change, and a change in the total tray holdup will affect the model through the values of M_i in equations (3. .1.c), (3. .2.c), (3. .3.c) and (3. .4.c).

The most important feature of the simulation is that it describes • the interactions between the fluid dynamic, heat transfer and mass transfer phenomena.

CHAPTER 5

DYNAMIC RESULTS - OSCILLATORY DISTURBANCES

This chapter presents an attempt to study the dynamic behavior of the column subject to oscillatory disturbances. The mathematical model has been found fairly adequate to represent the extractive column with disturbances such as step change and pulse change to feed variables. It was also mentioned that the model depends intimately upon the nature of the disturbance and the disturbed variable.

As a first step toward the identification technique using the black box concept, a test has been made with a square wave disturbance into thefeed flowrate variable. Due to a delay in the equipment arrival and the problems encountered when hooking up the control valves as described in chapter 2, a preliminary simulation study of the effect of the frequency and the amplitude which characterize the square wave, was performed on the feed flowrate variable, using the previous simulation program with some minor modifications. The feed flowrate was chosen as the disturbed variable since experimental work could be done later to confirm the simulation study. The reason for choosing a square wave type of disturbance was to overcome the non-linearity problem of control valves in the actual operating range.

The control valves were set up and a reasonable range was determined with the adjustment of the pressure in the feed tank as mentioned in

chapter 2. Due to lack of time, only one experimental run was performed with a square wave disturbance into the feed rate variable and the results have been compared with calculated results.

A brief survey about different approaches on unsteady operation of distillation columns is presented in the following section.

5.1 Literature review

Controlled-cycling operation of columns was generally understood as operation in which periods for dense phase to flow down and light phase to flow up are repeated alternately. Schrodt (36) explained qualitatively the improvement as followed: when two phases containing a transferable substance are brought in contact, the driving force for mass transfer is initially very high, than decreases to approach zero, close to equilibrium conditions; controlled-cycling conditions involve more fresh phases brought in contact, the time-average driving force is higher. McWhirter and Cannon (37) found an efficiency increase as high as 160 per cent on experimental distillation columns operating at total reflux.

A number of investigators such as Belter and Speaker (38) considered a cycle comprising periods for dense phase flow down the column, a coalescence stage with no flow, light phase flow up the column and again a coalescence stage with no flow. Efficiencies were obtained greater than 100 per cent and throughputs were 2 to 10 times greater than the steady-state case. Reference 35 is a useful source of information on controlled-cycling literature.

A more recent approach to unsteady-state operation of distillation process was analytically done by Matsubara et al. (39). Instead of dealing with cases in which a vapor flow period and a liquid flow period are repeated

alternately, Matsubara et al. conceive an optimal cycle of operations which, basically, consisted in taking off the top product only when the composition of the more volatile component in the condenser is high and the bottom product only when the composition in the reboiler is low. The numerical study was applied to a binary distillation column with negligible vapor holdup, constant liquid holdup on each stage, identical vapor rate throughout the column, identical liquid rate throughout the enriching or the stripping sections of the column, plates with 100 per cent efficiency and with variable condenser holdup and reboiler holdup. Matsubara et al. concluded that if the maximal admissible vapor rate is large, periodic control required a much smaller number of stages relative to steady control, to attain the same separation efficiency.

Friedly (34) has also mentioned the possible improvement of nonlinear processes when operated at unsteady-state conditions rather than at steady-state conditions. Lannus and Kershenbaum (33) noticed an improvement of the conversion in a tubular reactor when the inlet concentration was forced sinusoidally.

5.2 Simulation study

The extractive distillation column was simulated within the possible operating range determined by the work of D. O'Neill (10).

Any change in the feed flowrate involves a change in the fluid dynamics of the column. The same technique was used for a square wave input as was used for a step or pulse change in the feed flowrate. The flow increase was transmitted instantaneously down to the bottom output, changing the liquid flowrates coming out from the trays between the feed location and the bottom, therefore changing the holdup at the trays according

to the Francis weir relation.

Fig. 5.1.a. shows the overhead response when a square wave with a period of 0.1 hr., an amplitude of 12% was put into the feed flowrate; the response was observed during a period of time of 2 hr. and gradual stabilization to a higher steady-state level for the acetone overhead composition seemed to occur.

Simulated runs were also performed with three different periods of 0.2 hr., 0.1 hr., 0.04 hr. at a constant amplitude level of 12%, to study the effect of the period on the overhead composition response. Simulation time was limited to 1.4 hr. for economic reasons. From fig. 5.1.b, one can observe three trends:

(1) The acetone composition did not stay at the steady-state level but increased from it substantially. The methanol mole fraction also decreased significantly.

(2) The larger the period is the greater the deviation from the initial steady-state level.

(3) The gradient of the deviation with respect to the period of the wave is not constant. The jump from curve C to curve B is much larger than from curve B to curve A although the increase in the period is almost the same.

The effect of the amplitude of the wave on the overhead response was also investigated by doing runs at the same period value of 0.1 hr. and three different amplitudes of 12, 9, 3 per cent successively. Effects of amplitude on the overhead response is shown in Fig. 5.1.c. It can be seen that in the range of amplitude considered (3% - 12%), the effect of amplitude is very small. This would strongly suggest that most of the improved performance possible by cycling can be obtained with only a very small amplitude (3%) and there is little incentive to go to larger amplitude.



fig. 5.1.a: OVERHEAD COMPOSITION RESPONSES TO SQUARE WAVE DISTURBANCE IN FEED FLOWRATE

ELAPSED TIME SINCE CHANGE IN FEED RATE, HR.



ELAPSED TIME SINCE CHANGE IN FEEDRATE, Hr.



ELAPSED TIME SINCE CHANGE IN FEEDRATE, Hr.

% MOLE FRACTION

5.3 Comparison of experimental and simulated data for the case of a square wave disturbance to the feed flowrate

The experimental run was performed by disturbing the feed flowrate by a square wave of 12 per cent amplitude and at a period of .1 hr., through the control valve. As usual, the column was first brought to steady-state and then disturbed. Experimental conditions were presented in Appendix D. The transient response of the overhead composition was observed during a period of 1.5 hr., by taking samples every 10 minutes and analyzing with a gas chromatograph. The settings of the gas chromatograph and the integrator for the analysis can be found in Appendix E.

A comparison of the experimental data and the simulated data is presented in fig. 5.2. The following facts were observed:

(1) Experimentally, the column ceased operating at the initial steady-state level after the disturbance was put into the feed flowrate. The overhead acetone molar fraction increased in fair accordance with the simulated curve. A more detailed explanation of the response behavior by considering heat transfer, mass transfer, equilibrium, and fluid dynamics would be difficult.

(2) The experimental trend of the methanol composition agreed fairly well with the simulation during the initial period of about 1 hr., but the experiment showed that methanol almost disappeared from the overhead product at the end the response. Priming of the top trays was observed during this time. The overhead flowrate was decreasing significantly, and methanol could not reach the top due probably to tray weeping. However, water still appeared in the overhead product, which could be explained by the fact that



ELAPSED TIME SINCE INITIAL CHANGE MADE IN FEED FLOWRATE, MINUTES

water was fed very close to the top of the column. Reboiler level increased constantly and has to be readjusted, possibly due to the fact that most of the feed was flowing down.

The simulation did not give the same results in the last section of the response since the constant holdup assumption was used throughout the simulated period of time.

It is suspected that instability of the column would have occurred if the experiment had been continued further, due to the consequences of tray weeping.

5.4 Conclusion

This chapter presented a preliminary study on the response of the column due to controlled-cycling disturbances to feed flowrate variable. The preliminary simulated results were found interesting in the sense that the efficiency seemed to improve substantially when the feed flow-rate was cycled with a symmetric square oscillation.

The calculated effects of cycling the feed rate are, thus, at least qualitatively consistent with previous results found in the literature, which have shown considerable improvements in distillation column performance under oscillatory conditions. Furthermore, this work has been done on a multicomponent system as opposed to a binary system, usually mentioned in the literature. Further experimental work is needed on this effect, in particular on possible hydrodynamic instabilities introduced by the controlled-cycling of the feed flowrate (see chapter 6).

CHAPTER 6

CONCLUSIONS AND RECOMMENDATIONS FOR FUTURE WORK

A mathematical model was developed to describe the dynamics of a fifteen-stage extractive distillation column using a mixture of acetone methanol and water as the solvent. The unsteady-state component material balance equations were used together with the Enthalpic Wilson equation extended to a multicomponent mixture in the equilibrium calculations. Since the fluid dynamic response of the column was rapid relative to the temperature and composition responses, a quasi-steady state assumption was made in the simulations for the step and pulse changes in feed composition, feed rate and reflux ratio variables. The technique used when the column was subject to a sudden change in the flowrate (e.g. changes in the feed flowrate variable or reflux ratio variable) consisted in satisfying, at the beginning of a change, the total tray material balances by correcting instantaneously the liquid flowrates down the column and therefore, the holdups on the trays. The comparison with available experimental data was found fairly consistent within the possible operating range of the column and showed that a reasonable simulation has been obtained from the assumptions and strategy used.

The simulation of the column was developed inside a DYNSYS 2.1 environment which offered to the user many advantages and flexibilities, e.g. computer time savings, overhead work. Chapter 3 described the desirable features of such an approach.

The real time software developed by Jackson (11) was totally redeveloped to be used with the Generalized Operating System Executive created by J.P. Tremblay (22). The new package consisted of basic data logging tasks,

unit conversion tasks, inlet temperature control tasks and inlet flowrate control tasks. A routine to perform a square wave disturbance was also available.

The rearrangement of the feed and solvent system with the addition of the control valves allows a number of possibilities for future work. One of the areas briefly investigated in this work was the study on the dynamic response of the column (overhead composition) when it was subject to controlled-cycling disturbances. A preliminary simulation study showed that the overhead composition in acetone improved significantly when the feed flowrate was disturbed by a square wave. The higher the frequency of the oscillatory disturbance, the smaller was the improvement, in the range of conditions mentioned in chapter 5. This point should be interesting for further future investigation since Lannus and Kershenbaum (33) revealed an improvement in the conversion when the inlet concentration was forced sinusoidally, in the case of a tubular reactor. Also, as stated by Friedly (34), processes which are not strictly linear can sometimes be operated better in the unsteady state than in the steady state; periodic processing takes advantage of the fact that system non linearities can make the average system output different from the steady-state values. A distillation process is one of the many non-linear systems where the non-linearity comes essentially from equilibrium relations. The effect of the amplitude of the periodic wave was also investigated and was found negligible within the operating condition range of the column.

A comparison of the simulated results with the preliminary experimental data was made in chapter 5. A fairly good agreement occurred during the first half of the response, but divergences dominated at the later times. Signs of instability were noticed experimentally, such as top tray weeping,

reduction of overhead product flowrate. Since strongly-supported experimental results were not available, the answer to the question "is instability caused by controlled-cycling disturbances in feed flowrate?" will be worth investigating later on.

The following points should be considered for future work. (1) As discussed in Chapter 2, the feed and solvent arrangement is very inconvenient because of the long time spent in charging the tanks through the small hole at the top. Furthermore, no feed could be added while the column is operating. The storage system should be revamped to rectify the above difficulties. Also, it is desirable that the pressures in the feed tanks and in the solvent tank can be set independently. A method of setting the desired pressure at each run is discussed in Chapter 2.

(2) The software to control the input stream flowrates was developed. The constants of the controller could be determined by trial and error runs. Thereafter, the perturbation could be achieved with closed-loop control, in other words, just by perturbing the controller set point.

(3) A brief review of the literature showed that unsteady-state operation of non-linear processes could improve the performance compared to the steadystate operation. This fact is of great interest for the promising results obtained by the simulation study described in Chapter 5 of this work, when the feed flowrate was disturbed by a square oscillation, and the increase in the overhead acetone composition observed. However, more experimentation must be performed to better explain and analyze the instabilities which occurred during the preliminary investigation in this project.

(4) Since a reflux divider was used, the column, in reality, was subject to an oscillatory change in the reflux flowrate. In the simulation study on controlled-cycling changes in the feed flowrate, the frequency of the oscillation was found to affect significantly the response of the column. Therefore, the frequency of the reflux divider cycle, which also affected, in a rigorous sense, the fluid dynamics of the column, might have an effect on the response and should be checked in future work (reflux divider period \approx 5 sec; smallest period investigated \approx 150 sec) of course the same reflux ratio could be achieved with different frequencies for the reflux divider cycle.

(5) It was mentioned before that the accuracy needed for a simulation model depends upon the objectives of the simulation work. However, another potential direction for further investigation of extractive distillation modeling will be the study of the effects of various assumptions on the model accuracy. As an example, constant volumetric holdup at the stages could be relaxed and the percentage inaccuracy could be analyzed.

(6) A future requirement before any significant control study can be taken will be the implementation of an on-line automatic analyzer.

An outline of the progress of this long-term project is described below:

D. Jackson (11) : - Basic process instrumentation

- Interfacing with minicomputer (Real Time Operating System, original version)

D. O'Neill (10) : - steady-state operating conditions determination

- steady-state simulation program

 preliminary study of dynamic behaviour (experimental)

Present work

: - dynamic simulation for step and pulse changes in feed composition, feed flowrate and reflux ratio variables, using DYNSYS 2.1

- Addition of control flexibility for inlet streams

 Modification of real time software to be used with GOSEX Preliminary dynamic and experimental study on controlled-cycling effects on column (square oscillation in feed flowrate)

Future work

anticipated

- : Effect of reflux-divider cycling on the response of the column
 - Further investigation on the effect of a square-wave disturbance in the feed flowrate (experimental and simulation studies)
 - Basic assumptions checking in the model development for the extractive distillation column.

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APPENDIX A

CALIBRATION OF THE FEED AND SOLVENT LINES

Calibration curves have been obtained for the feed line and the solvent line. The pressure in the tanks was set at 15.6 psig.

For the feed line, within the possible limits of operation as determined by O'Neill (10), i.e. between 14 ml/min and 18 ml/min, the calibration curve of the feed flowrate vs the voltage read from the D.P cell, expressed in A/D units, could be described approximately a straight line, curve (2), whose equation is:

F = 0.1098 V - 15.7088 with F in ml/min V in A/D units

Curve (1) is the calibration curve of the valve action vs the flowrate, and could be approximated by:

 $P_{F} = 10.832 F^{2} - 316.492 F + 2500 with F in ml/min <math>P_{F}$ in D/A units

The same curves were obtained with the solvent line. Solvent flowrate versus the voltage from the DP cell, expressed in A/D units, is represented by curve (4) and the valve action versus the flowrate is represented by curve (3).

for curve (4) : S = 0.13331 V - 16.0023 for curve (3) : $P_S = -5.898 S^2 + 141.457 S - 440$ Fig. A-1 CALIBRATION CURVES FOR FEED STREAMLINE



fig. A-2 CALIBRATION CURVES FOR SOLVENT STREAMLINE



- with S in ml/min P_S in D/A units V in A/D units

A least squares fitting technique was used.

APPENDIX B

DYNSYS DATA SET

] Simulation Data

The simulation data begins with the data word "BEGIN". All cards before this are comments and are ignored by DYNSYS.

For example

IN/OUT	3.0	sets the maximum number of streams
		entering or leaving any module to
		three (default = <u>5</u>)
НМАХ	0.01	sets the maximum integration set size
		to 0.01 (default = 0.05)
TIME	3.0	sets the time of simulation to 3.0
		(default = 10.0)
LIBRARY	1.0	means that one new module will be de-
		fined immediately after this card.
		Most common modules such as CONT1,
		STIRl and VALVl are already in a
		library of module names stored inside
		DYNSYS.
DELAY	9.0	DELAY is the new module being defined.
		It will be accessed as SUBROUTINE
		TYPE9

Other possible simulation data words are:

6

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COMPS	Х	sets the number of components to X
		(default = 1)
DELTAT	X	sets the initial integration step
		size to X (default = 0.00001)
HMIN	X	sets the minimum integration step size
		to X (default = 0.000001)
TOLERANCE	Х	sets the error tolerance for the in-
		tegration to X (default = 0.001)
ORDER	х	sets the maximum permissible order of
		integration to X (default = 6)
NONSTIFF		causes the nonstiff option to be used
		(default = stiff option)
MINPIVOT	х	sets the minimum permissible pivot
		value for TRGB-TRGB2 to X
		(default = 0.000001)
PRINTING	х	causes output to be printed every X
		time steps (default = 1)
LINEPLOT		causes output to be printed as a line-
		plot (in this case additional data
		must be given after physical proper-
		ties data)
GRAPH		causes output to be printed as graph
		or graphs on CALCOMP plotter as well as
		lineplot INFPLOT must also be used

FUNCTION X causes X vectors of stream information to be read in as an input function. Stream data follows immediately.

The data word "PROCESS" signifies the beginning of the equipment data and thus the end of the simulation data.

.2 Equipment Data

The data word "PROCESS" signifies the beginning of the equipment data. Here data for the various modules is written in their order of execution.

CONT1 1.0 the first module to be executed is CONT1 , referred to as module #1 in the dynamic information flow diagram

After this, the entering and leaving stream numbers are given. A positive number indicates an input stream while a negative number indicates an output stream.

3.0 -4.0 stream 3.0 enters CONT1 and stream4 leaves it.

The following card gives the equipment parameters for the module. This is the data which describes the individual units.

3. 1500. 1000. 2. 3.

For example, the set point for the controller is the third parameter 1000. Up to five parameters may be given. If more are required, the data word "EXTRA" X must follow and the X extra parameters appear on the following data cards.

We continue in this way for all modules. The data word "END" signifies the end of the equipment data.

3 Stream Data

The material flows between equipment are represented by material streams. Information about each stream is stored in the following vector.

position 1 - stream number

- 2 flag
- 3 total mass flow
- 4 temperature
- 5 pressure
- 6 mass fraction of component 1
- 7 mass fraction of component 2,

etc.

The stream flag serves the dual purpose of identifying the stream type (flags with absolute value < 10 are material flows, those > 10 are information flows) and, if negative, suppresses printing of the stream vector in the output.

The user may choose his own system of units as long as he is consistent throughout. Usually English units are employed.

STREAMS 5. means that initial information data will be given for five streams. EXPLICIT means that the data will be given explicitly.

The following cards give the data. For the first stream

 1.
 1.
 1000.
 100.
 14.7

 1.
 .
 .
 .
 .
 .

the stream number is 1. The stream is a material flow (the flag is less than 10). The flow rate, temperature and pressure are 1000., 100. and 14.7 respectively. There is only one component so its mass fraction is 1.0.

Data is given for five streams. The data word "END" signifies the end of the stream data.

4 Physical Properties Data

PROPERTIES -1. means that the physical properties of water are assumed.

The user may provide his own data if desired. See the DYNSYS Manual for details.

The data word "END" signifies the end of the physical properties data and in this case the end of the data set.

APPENDIX C

Francis Weir Relation Used to Calculate the Liquid Tray Holdups

As mentioned in chapter 3, section 3.3.1 the assumption of constant volumetric holdup on the trays was made.

From the steady state simulation program, one can obtain the initial liquid molar flowrates leaving each stage. These flowrate values have been used with the Francis Weir relation to determine the initial volumetric holdups which will be kept constant throughout the dynamic simulation in this work. As described in reference 19 of Luyben, the liquid flowrate F_L is proportional to the weir length WL and to the overflow height h raised to a power of 1.5:

 $F_{L} = a (WL)(h)^{1.5}$

By rearranging

$$h = \left(\frac{1}{a} \frac{F_{L}}{WL}\right)^{2/3}$$

with a = 11988

 F_1 in ft³/hr, liquid flowrate

leaving the tray.

WL in ft, weir length.

h in ft, overflow height of tray

holdup.

The holdup height was then calculated:

$$H = h + WH$$

WH in ft, weir height

The volumetric holdup could be then obtained.

APPENDIX D

Run with Square Wave as Input Disturbance in Feed Flowrate

Operating conditions

Feed rate:	high level	=	18 ml/min
	low level	=.	16 m]/min
-	period of oscillation	=	6 min.
Feed composition:	60% mole acetone		
	40% mole methanol		
Solvent rate:	9.5 ml/min.		
Solvent composition:	pure water		
Reflux ratio:	1.772		
Solvent temperature:	65°C		8
Feed temperature:	40°C		
Feed variac setting:	15		
Solvent variac setting:	27		
Reboiler variac setting:	70		
APPENDIX E

Gas Chromatograph and Integrator Settings

Gas Chromatograph Varian 90-P3

Integrator Hewlett-Packward 3370B

-settings - noise suppresion - maximum
- slope sensitivity - up = 0.03
- down = 0.03



A M W

GM MOLES \times 10⁶ INDICATED ON SYRINGE



INTEGRATOR PEAK AREA

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-B.L. reset delay = 0.5 minutes

-area threshold = 100

-shoulder control - front - on

- rear = 100 mV

APPENDIX F

Dynamic Simulation Programs

A number of utility routines have been developed to assist the dynamic simulation in this work.

ENTHV:	this function calculates the enthalpy of a vapor stream
	at a specified temperature, pressure and composition
ENTHL:	this function calculates the enthalpy of a liquid stream
	at a specified temperature, pressure and composition,
	for the mixture of acetone-methanol-water, taking into
	account the heat of mixing of the components
ACTY:	this function calculates the coefficient of activity of
	each component in a mixture of three component using
	the enthalpic Wilson equation extended to multicomponent
	mixtures, as described in section 3.3.2.2.
KVAL:	this function calculates the activity of each component

in a mixture of components, as described in section 3.3.2.2.

The listings of the above utility programs and the main subroutine Type 1 are presented on the following pages.

SUBROUTINE TYPE1 * ******* * ¥ . ¥ . ¥ EXTRACTIVE DISTILLATION COLUMN . # . * BY D.T.TRANG , CHEMICAL ENGINEERING DEPARTMENT MCMASTER UNIVERSITY , OCTOBER 1975 THIS SUBROUTINE SIMULATES THE DYNAMICS OF AN EXTRACTIVE DISTILLATION COLUMN WITH THE FOLLOWING CHARACTERISTICS = 13 TRAYS+CONDENSER+REBOILER EXTRACTANT= WATER SOLUTION= ACETONE-METHANOL THIS SUBROUTINE IS DEVELOPED TO BE USED WITH THE DYNAMIC SIMULATION EXECUTIVE DYNSYS 2.1. (REFERENCE=BARNEY, J.R., PH.D. THESIS, ≠DYNAMIC_SIMULATION OF LARGE STIFF SYSTEMS IN A MODULAR SIMULATION FRAMEWORK≠, APRIL SIMULT AND TRIDAG ALGEBRAIEC EQUATION SOLVER= DYNSYS 2.1 DATA SET INPUT,OUTPUT LIQUID SOLVENT FEED STR LIQUID FEED STREAM OVERHEAD PRODUCT STREAM EQUIPMENT STREAM MP(IM,3) L MP(IM,4) L MP(IM,5) 0 FEED STREAM BOTTOM PRODUCT STREAM MP(IN,6) EQUIPMENT PARAMETERS EP(IM,1) TOTAL NUMBER OF TRAYS(INCLUDING REBOILER AND CON EP(IM,2) FEED TRAY EP(IM,3) SOLVENT TRAY EP(IM,4) PEFLUX RATIO EP(I,5) REBOILER DUTY KCAL/HP) EX(MP(IM,NMP+1)+I-1) STAGE I MUPPHREE EFFICIENCY FOP METHANOL EX(MP(IM,NMP+1)+I+19) STAGE I MUPPHREE EFFICIENCY FOP MAT EX(MP(IM,NMP+1)+I+19) CONDENSER HOLD UP (CM3) EX(MP(IM,NMP+1)+40) CONDENSER HOLD UP (CM3) EX(MP(IM,NMP+1)+41) REBOILER HOLD UP (CM3) EX(MP(IM,NMP+1)+42) WEIR HEIGHT (FT) EX(MP(IM,NMP+1)+44) WEIR LENGHT (FT) EX(MP(IM,NMP+1)+44) COLUMN DIAMETEP (FT)

STREAM DATA STREAM NUMBER STREAM FLAG TOTAL MOLAR FLOW (GMOL TEMPERATURE (DC) PRESSURE (MM HG) MOLE FRACTION ACETONE MOLE FRACTION METHANOL MOLE FRACTION WATER ,12) ,23) ,5678) ,8 , FLOW (GMOLES/HR) (DC) 3 7777 ***** SYSTEM STORAGE SPACE REAL HW COMMON/MAT/MP(5,8),EP(5,5),S(2,50,10),EX(100) COMMON/CON/IG,NCOMP,NC5,H,NE,NS,NPR,NPOL,TMAX,IORDER,NGRAPH COMMON/UNIT/IM,NMP COMMON/GEAR2/EPS,TIME,KFLAG,JSTART,NBVMAX,ICONV,ISTIFF COMMON/MODULE/IDERY,ITER,ITRI,HC,IPIVOT COMMON/SUBDI/A(50) COMMON/SUPERD/C(50) SUBROUTINE STORAGE SPACE VHUP(20),HUP1(20) ETHBAL(20),VDIF(20) Y(100),DERY(100) DIMENSION DIMENSION DIMENSION SECTION 1 INITIATION AND PARAMETER CALCULATION USE NEWTON-RAPHSON ITERATION WITH STIFF OPTION(ITER=1) ITER=1 IF(JSTAPT 0 .OR. IG .EQ. 1) GO TO 200 .NE. TRANSFER INFORMATION INTO SUBROUTINE SPACE REBOILER VOLUMETRIC HOLDUP CONDENSEP VOLUMETRIC HOLDUP WEIR HEIGHT WEIR LENGTH TRAY DIAMETER REFLUX RATIO TOTAL NUMBER OF TRAYS FEED TRAY SOLVENT TRAY VHUP(15) VHUP(1) WH WL DIA NFT

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VHUP(15) = EX(MP(IM.NMP+1)+41) VHUP(1) = EX(MP(IM.NMP+1)+40) WH=EX(MP(IM.NMP+1)+42) WL=EX(MP(IM.NMP+1)+43) DIA=EX(MP(IM.NMP+1)+43) RR=EP(IM.4) REBDT=EP(IM.5) NT=EP(IM.5) NT=EP(IM.2) NST=EP(IM.3) ADDITIONAL PARAMETERS LEVEL=1 MN=1 GYCLEF=.1 TIMP=FLOAT (MN) * CYCLEF CALCULATE THE AREA OF ONE TRAY TAREA=3.14/4.*((DIA-WL)*30.48)**2 NEQ IS THE NUMBER OF ORDINARY DIFFERENTIAL EQUATIONS NEQ=30 MC=50 CLEAR THE JACOBIAN STORAGE DO 99 I=1,MC A(I)=B(I)=C(I)=0. CONTINUE 99 100 C C C C CONTINUE CHECK TOTAL MASS BALANCE NTM1=NT-1 S(2,NT+20,3)=S(2,NTH1+20,3)-S(2,NT,3) DO 17 I=2,NTM1 DO J=NT+1-IBAL=S(2, J+20-1, 3)+S(2, J+1, 3)-S(2, J, 3)-S(2, J+2?, 3) SPECIAL TERMS FOR FEED AND SOLVENT TEAY IF(J.EQ.NFT)BAL=BAL+S(2,18,3) IF(J.EQ.NST)BAL=BAL+S(2,17,3) FLOW DOWN TO SATISFY STAGE TOTAL MASS BALANCE

. . D0 27 IK=J,NT S(2,IK+20,3)=S(2,IK+20,3)+BAL CONTINUE 27 17 C C C CONTINUE CHECK FOR CONDENSER S(2,16,3)=S(2,2,3)/(PR+1.) BAL=S(2,2,3)-S(2,16,3)-S(2,21,3) D0 37 IK=1,NT S(2,IK+20,3)=S(2,IK+20,3)+BAL CONTINUE 3000000000 CALCULATION OF INITIAL HOLD UP USING THE FRANCIS WEIR PELATION GIVEN THE INITIAL LIQUID FLOW RATE FROM EACH TRAY START CALCULATION FROM THE BOTTOM (EXCLUDING THE REBOILER AND T CONDENSER NTM2=NT-2 DO 1 II=1,NTM2 I=NT-II III=I+20 P=MW(IG, III) Q=DENS(IG, III) 000 CALCULATE THE OVERHEIGHT (IN FEET) =S(IG,III,3)*P/Q*3.5314*.00001 / /11988./WL)**.66667 FLOW H=(FLOW HH=H+WH CCCC CALCULATE THE MOLAR HOLD UP(IN GMOLES) VHUP(I)=HH*30.48*TAREA HUP1(I)=VHUP(I)*Q/P CONTINUE . 1 0000 WORKI G AREA FOR INTEGRATIONC TRANSFER THE VARIABLES INTO D0 5 I=1,NT II=I+20 MOLE FRACTION ACETONE NN=NT+1-I Y(NN)=S(IG,II,6) MOLE FRACTION METHANOL N=NN+NT Y(N)=S(IG,II,7) CONTINUE D0 30 I=3,8 S(1,17,I)=S(IG,17,I) S(1,18,I)=S(IG,18,I) CONTINUE С С 5 30 С c²⁰⁰ CONTINUE

0000		ADDITINAL SECTION FOR SPECIAL	DISTURBANCES	
č		PULSE CYCLING TEST SECTION		
7		IF(TIME .GE. TIMP)GO TO 7 GO TO 6 CONTINUE		
'		IF(LEVEL . EQ. 1) GO TO 9 S(1,18,3)=S(2,18,3)=16.96 LEVEL=1		•-
9		GO TO 10 CONTINUE		
10		S(1,18,3)=S(2,18,3)=15.04 LEVEL=0 CONTINUE		
		MN=MN+1 TIMP=FLOAT(MN)*CYCLEF		
6		GO TO 100 Continue		
C C C	CAL	SECTION 2 CULATION OF JACOBIAN MATRIX ON IF(IG.FQ.2) GO TO 300	CORRECTOR PASS	
CCCC				
C		AT NORMAL TPAY		
		N=NT+1-I IF(N-EQ-NFT) GO TO 13		a
		IF(N.EQ.NST) GO TO 14 ADD=0.		
13	,	GO TO 15 CONTINUE		
	÷	ADD=S(IG,18,3) GO_IO_15		
14		$\begin{array}{c} \text{CONTINUE} \\ \text{ADD=S(IG, 17, 3)} \\ \text{CONTINUE} \end{array}$		1
15	4	B(I) = B(15+I) = (S(IG, N, 3) - S(IG, N, 3))	-1+20,3)-S(IG,N+1,	3) +
12	-	C(I)=C(15+I)=S(IG,N-1+20,3)/HU CONTINUE	P1(N)	
C C		AT REBOILER		
C		B(1) = B(16) = -S(IG, 35, 3)	/ (VHUP(15)*DE	NS(IG,35)
r		C(1)=C(16)=S(IG,34,3)/(VHUP(15) * DENS(IG,35) / + W(I	G,35))
ŭ		AT CONDENSER		
16		B(15)=B(30)=(S(IG,2,3)-S(IG,15)	,3))/(VHUP(1)*DENS	(IG,21)/MW(IG,21)

С С 300 CONTINUE ITRI=1 000000000 SECTION 3 CALCULATION OF THIS SEGMENT 0 OF THE DERIVATIVES OF PROGRAM APPLIES FOR THE CASE OF 3 COMPONENT SYSTEM DO 20 I=2,NTM1 N=NT+1-I IF(N.EQ.NFT)GO TO IF(N.EQ.NST)GO TO FOR NORMAL TRAY 21 22 С ADD4=0 ADD5=C ADD5=C GO TO 23 CONTINUE 21 CONTINUE FOR FEED TRAY ADD4=S(IG,18,3) ADD5= S(IG,18,3)*S(IG,18,6) ADD6=S(IG,18,3)*S(IG,18,7) GO TO 23 CONTINUE FOR SOLNENT TRAY ADD4=S(IG,17,3) ADD5=S(IG,17,3)*S(IG,17,6) ADD6=S(IG,17,3)*S(IG,17,7) GO TO 23 CONTINUE VARIATION OF ACETONE COMPOSITION STATEMENT -- TRANSFER IGNORED L1=I 21 С c²² 23 C LL TRANSFEP STATEMENT -- TPANSFER IGNORED L1=I DERY(L1)=(-S(IG,N,3)*S(IG,N,6)+S(IG,N-1+20,3)*Y(L1+1)+ 1S(IG,N+1,3)*S(IG,N+1,5)-Y(L1)*(-S(IG,N,3)+S(IG,N-1+20,3)+S(IG 1,N+1,3)+ADD4)+ADD5)/HUP1(N) VARIATION OF METHANOL COMPOSITION L2=15+I DERY(L2)=(-S(IG,N,3)*S(IG,N,7)+S(IG,N-1+20,3)*Y(L2+1)+ 1S(IG,N+1,3)*S(IG,N+1,7)-Y(L2)*(-S(IG,N,3)+S(I3,N-1+20,3)+S(IG 1,N+1,3)+ADD4)+ADD6)/HUP1(N) GO TO 29 D CONTINUE C 20 VARIATION OF THE MOLE COMPOSITION AT THE REBOILEP -- TPANSFER IGNORED STATEMENT DERY(1) = (-S(IG, 15, 3)*S(IG, 15, 6)+S(IG, 34, 3)*Y(2)-S(IG, 35, 3)* 1Y(1))/ (VHUP(15)*DENS(IG, 35)/MW(IG, 35)) DERY(16) = (-S(IG, 15, 3)*S(IG, 15, 7)+S(IG, 34, 3)*Y(17)-S(IG, 35, 3)* 1Y(16))/(VHUP(15)*DENS(IG, 35)/MW(IG, 35)) C C VARIATION OF THE MOLE COMPOSITION AT THE CONDENSER

C DERY(15)=(S(IG,2,3)*(S(IG,2,6)-Y(15)))/(VHUP(15)* 1DENS(IG,21)/MW(IG,21)) DERY(30)=(S(IG,2,3)*(S(IG,2,7)-Y(30)))/(VHUP(15)* 1DENS(IG,21)/MW(IG,21)) CCCC INTEGRATE CALL DIFSUB(NEQ, Y, DERY) IF(IDERY .NE. 0) GO TO 300 CCCCCCCCC0997 SECTION 5 OUTPUT MANIPULATIONS MANIPULATION 1 TRANSFER VARIABLES INTO SYSTEM STRORAGE CONTINUE DO 8 I=1,NT II=I+20 NN=NT+1-I S(1,II,6)=Y(NN) N=NN+NT S(1,II,7)=Y(N) S(1,II,8)=1.-Y(NN)-Y(N) CONTINUE DO 31 I=1,35 DO 32 J=4,5 S(1,I,J)=S(IG,I,J) CONTINUE CONTINUE 8 321 31 C C C C C C MANIPULATION 2 CALCULATION OF THE MOLECULAR HOLD UP DO 43 II=1,NTM2 I=NT-II III=I+20 P=MW(1,III) Q=DENS(1,III) HUP1(I)=VHUP(I)*Q/P CONTINUE 4000000 MANIPULATION 3 CALCULATION OF THE COMPOSITION OF THE VAPOR LEAVING EACH TPAY AT THE REBOILER III=NT II=NT+20 DUM=EQUIL(1,II,III) TRUE SEPARATION AT REBOILEP CC S(1 ,III,7)=EX(MP(IM,NMP+1)+NT-1)*S(1 ,III,7)+S(1 ,II,7)*(1.-1EX(MP(IH,NMP+1)+NT-1))

ç	S(1,III,8)=EX(MP(IM,NHP+1)+NT+19)*S(1,III,8)+S(1,II,8)*(1 1EX(HP(IM,NMP+1)+NT+19)) S(1,III,6)=1S(1,III,7)-S(1,III,8)
C C	CALCULATION FOR ALL OTHER TRAYS(INCLUDING THE CONDENSER) DO 2 I=2,NT III=NT-I+1 II=III+20
C	DUM=EQUIL(1,II,II)
C	IRUE SEPARATION AT EACH TRAY
C	<pre>S(1, fill, 0)=2x(MP(IM, NMP+1) + III+19) = S(1, fill, 3) + S(1, fill+1, 2) = 1(1Ex(MP(IM, NMP+1) + III+19)) S(1, III, 7) = Ex(MP(IM, NMP+1) + III-1) * S(1, III, 7) + S(1, fill+1, 7) = 1(1Ex(MP(IM, NMP+1) + III-1)) S(1, III, 6) = 1S(1, III, 7) - S(1, III, 8) 2 GONTINUE</pre>
50 00 26	MANIPULATION 4 SET SAME COMPOSITION AT OVERHEAD AND REFLUX STREAMS DO 26 I=1,NCOMP III=I+5 S(1,16,III)=S(1,21,III) CONTINUE
	SET SAME TEMPERATURE S(1,15,4)=S(1,21,4)
000000	MANIPULATION 5 CALCULATE THE VAPOR FLOWRATES AND LIQUID FLOWRATE USING THE ENERGY EQUATION
C	<pre>FOR THE REBOILER III=NT II=NT+20 ETHBAL(III)=REBDT+S(IG,II-1,3)*ENTHL(1,II-1)-S(IG,II,3) 1*ENTHL(1,II)-S(IG,III,3)*ENTHV(1,III) VDIF(III)=ETHBAL(III)/(E'NTHV(1,III)-ENTHL(1,II))</pre>
000000	CALCULATION FOR ALL TRAYS EXCEPT THE CONDENSER INITIATE THE CALCULATION
U	HLOUT=ENTHL(1,NT+20-1) HVIN=ENTHV(1,NT) DO 3 I=2,NTM1 III=NT+1-I II=III+20 HLIN=ENTHL(1,II-1)

.

HVOUT=ENTHV(1, III) IF(III.EQ.NFT) GO TO 87 IF(III.EQ.NFT) GO TO 81 NORMAL TRAY 8D CONTINUE ADD=0. GO TO 82 FOR FEED AND SOLVENT TRAYS CONTINUE ADD=S(1, 18, 3)*ENTHL(1, 18) GO TO 82 81 CONTINUE ADD=S(1, 17, 3)*ENTHL(1, 17) GO TO 82 82 CONTINUE P. STATEMENT -- TRANSFER IGNORED ETHBAL(III)=S(IG,III+1,3)*HVIN+S(IG,II-1,3)*HLIN-S(IG,II,3) 1*HLOUT-S(IG,II,3)*HVOUT +ADD VDIF(III)=ETHBAL(III)/(HVOUT-HLOUT) REINITIATE FOR NEXT TRAY CALCULATION

```
HVIN=HVOUT
HLOUT=HLIN
GC VAPOUR FLOWPATES
C VAPOUR FLOWPATES
C TVAP=0.
BO 33 I=1,NTM1
III=NT+1-I
TVAP=TVAP+VDIF(III)
S(1,III,3)=S(IG,III,3)+TVAP
CONTINUE
C FOR THE CONDENSER
S(1,16,3)=S(1,2,3)/(RR+1.)
TLIQ=S(1,2,3)-S(1,16,3)-S(IG,21,3)
S(1,21,3)=S(1,21,3)+TLIQ
TVAP=TLIQ
DO 34 I=2,NT
TVAP=TVAP-VDIF(I)
S(1,I+2D,3)=S(IG,I+2C,3)+TVAP
CONTINUE
C C FOR THOUSE
C C FOR THE CONDENSER
S(1,21,3)=S(IG,I+2C,3)+TVAP
CONTINUE
```

С

C C C 87

C

ULL TRANSFER

RETURN 105 CONTINUE PRINT 500 500 FORMAT(10X, #****ERROR AT REBOILER*****# STOP END

FUNCTION ACTY(IG, IS)

C 8888 Ĉ

888

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THIS PROGRAM CALCULATES THE COEFFICIENT OF ACTIVITY OF EACH
COMPONENT IN A ACETONE-METHANOL-WATER MIXTURE,
USED WITH DYNSYS 2.1
               COMMON/MAT/MP(5,8), EP(5,5), S(2,50,10), EX(100)

COMMON/PTAB/IGFLAG, PP(10,20)

COMMON /CON/ IU, NCOMP, NC5, H, NE, NS, NPR, NPOL, TMAX, IORDER, NGPAPH

COMMON/KVALU/HKK(10)

DIMENSION A(3), B(3), C(3), AA(3), B9(3), CC(3), X(3), GAM(3,3)

, ACT(3), LAM(3,3), Y(3), V(3), VV(3), Y2(3)

DIMENSION ALP(3,3), ALP1(3,3,3), SUMGX(3)

REAL LAM

DATA AA/55, 8622, 21, 795, 22, 8876/ PP(8, 45555-7, 4, 05075, 7, 7, 6)
             DATA AA/56.8622,21.795,22.8876/,BB/8.4555E-3,4.9507E-2,-3.642E-2/,

1 CC/1.65E-4,5.68E-5,6.856E-5/

LAM(1,2) =-216.966

LAM(2,1) =506.093

LAM(1.3) =-4.56.093
               LAM(2,1)=506.093
LAM(1,3)=-436.468
LAM(3,1)=1185.436
LAM(2,3)=-208.7681
LAM(3,2)=571.3078
ENTHALPIC WILSON EQUATION TO PREDICT VAPOUR MOL FRACTION IN EQUIBRIUM
WITH GIVEN LIQUID CONCENTRATION
BINARY COEFFICIENTS FOR WILSON EQUATION, VAPOUR PRESSURE ANTOINE
COEFFICIENTS AND LIQUID MOLAR VOLUMES FROM
ACETONE -1. METHANOL -2.WATER -3
                ACETONE -1, METHANOL -
TK=S(IG,IS,4)+PP(1,20)
                                                METHANOL
                                                                        -2, WATER
                ACTY=C.
                CALCULATE THE GAMMA≠S BY (15-B)
                00
                          20 J=1,3
               UC 2U J=1,3

GAM(J,J)=1.

DO 10 I=1,3

IF(I.EQ.J) GO TO 19

FC=EXP(-LAM(I,J)/(1.987*TK))

V(J)=AA(J)+(BB(J)+CC(J)*T()*TK)

VV(I)=AA(I)+(BB(I)+CC(I)*TK)*TK

GAM(I,J)=V(J)/VV(I)*FC

CONTINUE

CONTINUE
       10 CONTINUE
20 CONTINUE
                READ IN THE PROPERTIES OF THE STREAMS
               D0_70_I=6,NC5
                N=I-5
               X(N) =S(IG,IS,I)
SONTINUE
DO 107 I=1,3
SUMGX(I)=0.0
     70
               D0 101 IP=1,3
SUMGX(I)=SUMGX(I)+GAM(I,IP)*X(IP)
               DO
     101
               CONTINUE
DO 102 I=1,3
DO 102 J=1,3
1.07
     102 ALP(I, J) =1.0+X(I)*(1.0-1.0/SUMGX(I)-GAM(J,I)/SUMGX(J))
               DO 106 I=1,3
DO 105 J=1,3
IF(I.EQ.J) GO TO 105
DO 103 M=1,3
               IF(M.EQ.I.OR. M.EQ.J) GO TO 103
ALP1(I,J,M)=1.D-GAM(J,I)/SUMGX(J)-GAM(M,I)/SUMGX(M)
                                                                            GO TO 103
 :103
:105
               CONTINUE
CONTINUE
  :106
               CONTINUE
               DO 60 I=1,3
               SUM1=0.0
```

```
DO 30 J=1,3
SUM1=SUM1+X(J) *ALOG(GAM(I,J) *GAM(J,I)) *ALP(I,J)/(SUMGX(I) *SU™GX
(J))
CONTINUE
         1
   30
          SUM2=0.0

D0 50 M=1,3

IF(M.EQ.I) G0 T0 50

D0 55 J=1,3

IF(J.EQ.I .OR.J.EQ.M) G0 T0 55

SUM2=SUM2+X(J) *X(M)*ALOG(GAM(J,M)*GAM(M,J))*ALP1(I,J,M)/

(SUMGX(J)*SUMGX(M))

CONTINUE
         1
           CONTINUE
   55
          CONTINUE
PP(I,9)=EXP(-SUM1-SUM2)
CONTINUE
    50
   60
           RETURN
            REAL FUNCTION KVAL(IG, IL, IV)
            THIS PROGRAM CALCULATES THE ACTIVITY OF EACH COMPONENT IN A MIXTURE
           COMMON/MAT/MP(5,8),EP(5,5),S(2,50,10),EX(100)
COMMON /CON/ IU,NCOMP,NC5,H,NE,NS,NPR,NPOL,TMAX,IORDER,NGRAPH
COMMON/PTAB/IGFLAG,PP(10,20)
COMMON/KVALU/HKK(10)
DUMMY STATEMENT
            KVAL=0.0
           KVAL=0.0
T=S(IG,IL,4)+PP(1,20)
P=S(IG,IL,5)
DUM=ACTY(IG,IL)
DO 1 I=1,NCOMP
HKK(I)=10.**(PP(I,6)-PP(I,7)/(PP(I,8)+T-273.))*PP(I,9)/P
CONTINUE
RETURN
FEIURN
FEIURN
FEIURN
   1
            END
                                                                                                                                                      ETV
            FUNCTION ENTHV(IG, IS)
           THIS PROGRAM CALCULATES THE ENTHALPY OF A VAPOR MIXTURE,
USED WITH DYNSYS 2.1
           COMMON/MAT/MP(5,8),EP(5,5),S(2,50,10),EX(100)
COMMON/CON/IU,NCOMP,NC5,H,NE,NS,NPR,NPOL,TMAX,IORDER,NGRAPH
COMMON/PIAB/IGFLAG,PP(10,20)
           COMMON/PTAB/10: ----

T=S(IG,IS,4)

HV=0.

DO 10 I=1,NCOMP

HV=HV+S(IG,IS,I+5)

HV=HV+S(IG,IS,I+5)

+PP(I,12))*T+PP(I,14))*T+PP(I,13))*T

+PP(I,12))*T+PP(I,11))*T+PP(I,10))
110
          K
```

6666

C

C

0000

```
THIS PROGRAM CALCULATES THE ENTHALPY OF A LIOUID MIXTURE
OF ACETONE-HETHANOL-WATER, USED WITH DYNSYS 2:1
COMMON/HAT/MP(5,8), EP(5,5), S(2,50,10), EX(100)
COMMON/TAB/IGFLAG, PP(10,20)
DIMENSION X(3)
HACEL(1)=30.6E-3+T
HMETL(T)=19.74E-3+T
HWATL(T)=18.016E-3+T
FORMULATION OF HEAT OF MIXING
THE PARAMETERS ARE GIVEN BY A LEAST SQUARE FITTING
HMIXWA(XC)=-6.00506E-3+D.368503*XC-2.1499*XC**2+5.3965F*XC**3
1-3.50732*XC**L
HMIXWA(XC)=2.01772E-3-P.753418*XC+9.94409*XC**2-0.192395*XC**3
HMIXWA(XC)=6.05874E-3+0.171893*XC+9.646148*XC**2-0.813034*XC**3
SET THE NECESSARY VARIABLES FROM THE STREAMS
T=S(IG,IS,4)
D0 19 1=1,3
II=1+5
X(I)=S(IG,IS,II)
CONTINUE
ENTHL=HACEL(T)*X(1)+HMETL(T)*X(2)+HWATL(T)*X(3)
IF(X(1)+E(2)
X1A=X(2)/X1B
X2B=X(3)+X(2)
X2A=X(3)/X2B
ENTHL=ENTHL-X1B*HMIXMA(X1A)-X2B*HMIXHM(X2A)-X3P*HMIXWA(X3A)
CONTINUE
ENTHUE
ENTHL=ENTHL-X1B*HMIXMA(X1A)-X2B*HMIXHM(X2A)-X3P*HMIXWA(X3A)
CONTINUE
ENTHUE
```

5.7

FUNCTION ENTHL (IG, IS)

10

APPENDIX G

Distillation Control Programs

The control software of the extractive distillation column has been developed under the GOSEX created by J.P. Tremblay (22). The minicomputer NOVA/2 in the Chemical Engineering computer room was equipped with necessary peripheral devices (A/D converter, D/A converter, relays, contact sense, one operator console and one data logging line printer) and could be used to run the column.

The user must provide:

- (1) a symbol table to facilitate duties as parameter change, variable logging...(SYMBOL OF TAPE #2 OF 2)
- (2) a disk file containing user messages (MESSG.TR)
- (3) user control programs (INFCE, FSCON, SWAVE, SSAMP)

The listings are presented on the following pages with explanations.

The listings of GOSEX parameter tape, symbol table and executive message file (see reference 22) are also presented.

A loading map could be found at the end of Appendix G, and the loading procedure is detailed in reference 22.

1801

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1101 101		
	TITL INFCE	
	; INTERFACE TO THE GO	BEX EXECUTIVE
	; DECEMBER, 1975	
	I MUMASTER UNIVERSITY	,
	; CHEMICAL ENGINEERIN	15 DEPARTMENT
	. ENT OVLAY, ALARM, DAT	IJM, RELAY, SENSE
	. ENT SCAN, CLOOP, SUPP	S, TASKS, UDEF
	. EXTN FEEFT, SOLFT	
	EXTN SOLFM, SOLVE, FE	EFM, FEEFB.
	EXTN FEETC, SOLTC	
	EXIN IRAF, SSAMP	
	. EXID TIMER	
000001	. TXTM 1	
	. NREL	
	; OVERLAY FILENAME DEFINITION	i
001000021	"OVLAY: . +1*2	
	, TXT @DISTIL OL@	
011'042111		
021051524		
B 1044514		4
041027117		
5.046000		
	JALARM TABLE DEFINITION	
6 000010	ALARM: 10	;# OF ALARMS IN TABLE
000010	. BLN 10	
000012	DATUM: BLK ATODS	; A/D CHANNELS
000012	RELAY: BLK RLAYS	; RELAYS
381177777	SENSE: TRAP	
1411000043·	TRAP	
55°000044	TRAP	
16 ⁻¹ 000045	TBAP	1.4.1
771000046	TRAP	
5001000047	TRAP	
5H1000050-	TRAP	
522100005t+	TRAP	
53F1000052+	TRAP	
54F10000531	TRAP	
	; MULTIFLEXER SCAN RATE	
511000002	SCAN 2	; 2 TIMES PER SECOND
	FID CONTROL LOOP DEFINITION	
56-177777	CLOOP: SOLFM	; SOLVENT LOOP
57'' 177777	SOLVE	
01177777	FEEFM	;FEED LOOP
51 - 177777	FEEFB	
21100000	160	
	USER SHUTDOWN ROUTINE	
3: 054407	SUFRS STA 3 . +7	
4-126420	SUBZ 1 1	;ZERO AC1
51044757	STA I RELAY+6	; TURN OFF FEED HEATER
6 044750	STA I BELAY+5	; TURN OFF SOLVENT HEATER
57 ° 02400 1 4		WAIT FOR RELAYS TO BO SET
7014004017		
111010400		
7::0000000		
2 000000		
- UU2/7/	UNE 2 F1	
12 1777	TASKS, FEETO	FEED TEMP MONITOR
- 1/////	(ASKS) PEEUU	

THEOR

.

TINFCE
5-000110
51000100
7~100000
0-177777
4~000110
2 000110
5-100000
4-000056
5-'000110
6~000120
7~100000
0~000060*
1-1000110
211000130
3~100000
4 177777
5~000110
6:000140
71/100000
01177777
1 '000110
2:1000150
3:1100000
41-177777
51000110
6,1000160
27**100000
001100000

1801 120

SOLVENT RATE MONITOR

; SOLVENT TEMP MONITOR

FRIDRITY

; IDENTIFICATION

FEED RATE MONITOR

1.12.14.14.14

SAMPLING MONITOR (OVERHEAD AND BOTTOM)

; SOUARE WAVE IN FEED RATE ROUTINE

; INFUT FLOWRATES SETTING ROUTINE

BIL'000000 UDEF:

0 . END

2	2	1	
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ί,	_		
١,	ſ	1	
l	1		

121 ENG UNITS ヨピントウドヨーンコー FOR FEED RATE AND OF DATA INTO ENG UNITS CHANNEL CHANNEL UNITS, READ-STINU RATE **DTNI** SOLVENT 나이니 ENG AND CONVERT DATA TO ENG ALSO CONVERT 3 OTHER TEP INGS TO ENG. UNITS SOLVENT FEED DATA CONTROL PERIOD CONTROL PERIOD SET POINT DEF CONTROL OF FE 5 CONTROL FOR SI CONVERSION OF LNIOL にど田之にたな山田口 CONTROL 5 T SOLFM, SOLVB, FEEFB, FEEFM F FEETC, SOLTC F FNOFP, FISET, ADFTE, FTEMP T SNOFP, STSET, ADSTE, STEMP F FFLOW, OPEAF, FFCON TABNS EXTN CONTL, BUHD, BNURS, MTPLY 100 PID CONTROL 47 GN/JFF FRO/ND WILLOW, OPENA, MULLON ENGINEERING FNOFT 119611 SNUFF 51521 PROGRAM HAS AND UNIVERSITY 014 AND LNEPH, LNURG FEETB, SOLTB DUTES, DUTEF 1975 CONVE, ADT FISCON TIMER DECEMBER, , MCMASTER TITL FSC FEETC SOLTC SOLFM FEETN NIXI EXTD NREL -NH ENT ENT ENT ENT トンヨ ENT ENT EN-1

...

HALLING R DONTROL ヨピリーマドヨムとヨー FEED

(IN A/D UNITS) してい READING FUIDT REALING DECIM した日に CHORE'S n Kult REFERENCE HOLINS RELAY WORD 見日に WALT PERIOD RAW FEED POINT TEMPT LEXEL TEED RELAY FEED T : A/D-A/D 1-10 L STORE INSTALL J. SKIT 1.661 1.661 1.661 1351 . FEETB-1 O SNC 1201100 CIVUTU9 AUFTE ACONE FISET GRILWF AUFTE **LLONE** ロシビロ NOTE QAUUAD USL: N 14 ---. SVETM DELAY (9,19) 0 0 0 -------N 0 51JB2# ----CDA STA LUA LUA SUB STA 1000 LUA E UA はこの 104 2019 LUA 110 D. FEETC 02/026427 03/022427 01/054016 00.034434 004000 05-044420 07/00/424 111-045004 22-020414 3/030414 15-122433 (8.152400 06-020447 10:030437 115520. 51 114250.21 00.024403 10900 002010

i dud

FSCON 23'000764 FNOFP: 500 24'000755 JMP FEETC+2

251000000 ADFTE	0
261000000 FTSET.	0
271001000 FHSW	1000
000003 RLWF:	GADD RELAY, 3
000004 DTUM4-	GADO DATUM, 4
000005 DTUM5:	GADD DATUM, 5
331177777 AQUAD	ØGAD
3-410000351 ⁻¹	FEETB
000012 FEETB:	. BLK 10.
17 000050 AENT:	ENTAB
5101000000 ENTAB	0
51.000000	0
5:2-000000	0
531000000 STEMP:	0
541000000 FTEMP	0
5510000561ACONB:	CONVE
5%175710 CONVE:	-1080.
57 076272	+31930.
507000000	+0

; SOLVENT TEMPERATURE ROUTINE

511-034470	SOLTC:	LDA 3 SOLTE-L
2 054016		STA 3 USP
3026447		LDA I CUTUMB
4 022746		LDA O GUTUMS
5106400		SUB O 1
6 044445		STA 1 ADSTE
W1020766		LDA O ACONB
701006743		JSR @AQUAD
11-030756		LDA 2 AENT
21045003		STA 1 3 2
38-020441		LDA O STSET
44.030442		LDA 2 SHSW
551024436		LDA I ADSTE
766-122433		SUBZ# 1 0 SNC
77-152400		SUB 2 2
001052435		STA Z CRLNS
01.1024436		LDA I CB
0211044436		STA L CNT
38°024441		LDA I AADT
)41.044441		STA I ADTI
51 024742		LDA L AENT
6-044435		STA L AENTI
7**024432		LDA I DITUMO
011044432		STA 1 CNT1
1/026431		LDA 1 CONTI
2: 022720		LDA O RDTUMS
31 106400		SUB O 1
4-1046431		STA 1 GADTI
51 020740		LDA O ACONB
0 006715		JSR @ADUAD
1 046424		STA I CAENII
v 010423		ISZ AENTI

; CONVERSION

IBIT 6

; TABLE IN ENG. UNITS

; INSTALL RELAY ; CONVERSION OF OTHER TEMP

CONTROL ACTION

; A/D-A/D REFERENCE ; STORE RESULT IN A/D TABLE 1001

						1803
FRIDAN						,
2.1 010424		ISZ A671				123
2.2 010420						
1.5 U14415 14:000745						
14 000765						
2.3 924493 3.4 004017		LUA I SINGAP Ivetn		THE WALLFE	01170	
271010400		DE SY				
30.0007744	ENGEE	500				
311000732						
000003	DTUME	GADO DATUMA 3				
313 1000000	ADSTE:	0				
5-4-0000000	STSET:	Ō.				
000002	ALWS:	. GADO RELAY, 2				
\$661002000	SHEW	2000		/ BIT 5		
317-000003	C.3 1	З				
4001000000	ONT:	0				
000000	DTUMO	GADE BATUM, O				
412-1000000	CNT1:	0	4			
4031000000	AENT1:	Ŏ.		•		
44.000146	AADT:	AUT				
45 000000 17 1000000	ADT1:	0				
*# 000000	ALT.	0		TABLE OF TE	MP IN AND UNI	75
4 <i>M -</i> 000000		0				2
50-000000	.•					
0000132						
000012	SULTE	. BER. IV		16		
	FID CO	NTROLLER FOR SOL	VENT FLOWR	ATE		
549°034435	SOLFM:	LDA 3 SOLVE-1		GET CONTROLLE	ER BLPCK ADDRE	55
657054016	4	STA 3 USP				
56° 023427		LDA O @VABLE 3		; GET RAW DATA		
57 041423		STA O SMOTH 3		;STORE		
700-1777777		CONTL		; CALL CONTROLI	LER	
	; CONVER	STON TO ENG UND	TS FOR OUT	FUT.		
711-021450		LDA O SPCON 3		ACONVERT INFUT	T VARIABLE	
72 025423		LDA L SMOTH 3	÷			
130 007451 741 0 1 5 4 1 4		JSR @SPAZE 3	4		3	
74# 04441/6 (55) 051 455		STA I SELOW		. A AND THE ALLER	T ACTICN	
1.30° 0214.34		LUA O MULINA S		JUNVER I DUTPI	צונון ונוא ונ	
177-1007422		LUA I DUTET S				
0074.J.J 001044415		CTA / ADESC	× 1			
)11-075447		INA I DELAV R				
221030001	a	IDA 7 TIMER				
1301177777						
141-101004		NOV 0 0 575				
151 126000		ADC I I				
15: 006017		SYSTM				
17**010400		DELAY				
100 0000000		0				
11.1000757		JMP SOLFM-4				
1211000000	SFLOW	0		1		
12: 000000	OPERS	Ŭ				
1411177502	SECON	-190		;A≈10≈≈5		
15/1003136		± 1650		, B*10**3		
D 175210		-1400	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	10*10		
000000	OUTES	Ō.	, TRUE A	*10 **3		
000000		Ú	, TALIE BA	HÚ - Agent Magneter	X1.7	
	FLORITADI	LLER DATA BLOCK	FUR SULVEN	RAIE MUNINHI	1/13	

1803FSLON 211/0002227 SOLVE 124 000026 SOLVE: BLK 22 50 000000 O_{-} FILTER 000006 GADD DATION, A ADDRESS OF INFUT DATA 52 000000 Ō. ; INPUT HIGH ALARM LIMIT 55 000000 Ō. FINEUT LOW ALARM LIMIT 54 000000 Q. HI ALARM MESSAGE NUMBER 551000001 Ł LO ALARM MESSAGE NUMBER 56 000002 $\mathbf{\tilde{z}}$ ITIME INTER FOR SUCC ALARM DISFL 577 '000000 Ũ. ISET POINT IN AVE UNITS 000000000 Ō. ; KP 511 (0000000) Ō. ; KFD 9000001至6 0 FKF I 63-000000 Ō 1 SCALE 54 000001 Ł ; DELAY 6510000000 0 ; DIRECT OR CASCADE CONTROL 661000000 0 ; D/A CHANNEL NUMBER 577000000 0 ; UPPER LIMIT ON OUTPUT ACTION 7000000 Ō. LOWER LIMIT ON OUTFUT ACTION 711/000000 Ō ; MEAN VALUE OF OUTPUT ACTION 72:000214 SECON ; ADDRESS OF CONSTANT BLOCK FOR CONV. 73-0000334 ; ADDRESS OF CONV. ROUTINE DUAD 74-177777 **DNVRS** ; INVERSE CONVERSION 755'000144 100. 7651000217 DUTES 777-177777 LNEAR GO:177777 LNVRS 0111000145 101 022-000146 102 ; PID CONTROLLER FOR FEED FLOWRATE 031034435 FEEFM: LDA 3 FEEFB-1 FGET CONTROLLER BLOCK ADDRESS 041054016 STA 3 USP 1501023427 LDA O EVRBLE 3 ; GET RAW DATA 66-041423 STA O SMOTH 3 77 000170 CONTL FCALL CONTROLLER ; CONVERSION TO ENG. UNITS FOR OUTPUT 00.021450 LDA O SPECON 3 ; CONVERT INPUT VARIABLE IL 025423 LDA I SMOTH B 221007451 JSR @SPAZE 3 38'044416 STA 1 FFLOW 44.021454 LDA O MNCON 3 ; CONVERT OUTPUT ACTION 55 025422 LDA 1 OUTPT 3 JSR @MNAZE 3 6 007455 71.044413 STA 1 DPERF 0/1025442 LDA L DELAY 3 1.0300015 LDA 2 TIMER 2110002031 MTPLY 31101004 MOV 0 0 SZR 411126000 ADC 1 1 51006017 SYSTM 6/010400 DELAY 7.000000 0 01 000757 JMP FEEFM+4 1. 000000 FFLOW 0 21'000000 DEERE-0 SF176650 FECON: ; A*10**5 -600 41/005132 ; B*10*3 2650 51176042 -99010810 6, 000000 OUTBE ; TRUE A \$10 **3 0

125 FRAGN T-10000000 O. ; TRUE & #10 CONTROLLER DATA BLOCK FOR FEEDRATE ; MONITORING FEEFB 011000341 BLN 22 000024 FEEFS: 7 000000 FILTER O. GADD DATUM, 7 ADDRESS OF INFUT DATA 000007 11-1000000 ; INFUT HIGH ALARM LIMIT Ō ; INFUT LOW ALARM LIMIT 1211000000 Ō 3-1000002 2 ; HI ALARM MESSAGE NUMBER ; LO ALARM MESSAGE NUMBER 14-1000003 з TIME INTER FOR SUCC ALARM DISP 75-1000002 2 SET POINT IN AVD UNITS 76-1000000 Ō) KP 77.000000 0 001000000 Ō ; KPD FRET 0000001110 0 ; SCALE 02:1000000 Ō ; DELAY 03:1000001 L ; DIRECT OR CASCADE CONTROL 04/000000 Ō , D/A CHANNEL NUMBER 05H000001 1 ; UPPER LINIT ON OUTPUT ACTION B-1000000 0 LOWER LIMIT ON OUTPUT ACTION 0711000000 \mathbf{O} ; MEAN VALUE OF OUTPUT ACTION 1001000000 Õ ; ADDRESS OF CONSTANT BLOCK FOR CONV 11.10003333 FELON 12:100027:3 ; ADDRESS OF CONV ROUTINE QUAD. ; INVERSE CONV 1351000274 **DNVRS** 141 000144 100. 155:000336 DUTEF 165 000277 LNEAR 177 1000/3001 LNVRS 2001000145 101. 211-000147 103.

END

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TITL SSAMP ; ROUTINE TO ACKNOWLEDGE CHANGES OCCURRED AT PROCESS SITE AND TO PRINT OUT AN APPROPRIATE MESSAGE ; RELAY BIT DAND CONTACT SENSE BIT O FOR OVERHEAD CHANGE FRELAY BIT I AND CONTACT SENSE BIT I FOR BOTTOM CHANGE ; DECEMBER, 1975 ; MCMASTER UNIVERSITY ; CHEMICAL ENGINEERING DEFARTMENT EXTN REC ENT SSAMP, TRAP NREL 07034443 SSAMP: LDA 3 SSAMB-1 STA 3 USP SUB 0 0 STA O TRAP LDA O TRAP-1 REC LDA O MASK 7~123420 ANDZ 1 0 ; MASK UNNANTED BITS SUBZE 2 2 FUT I IN AC2 0≅152520 SKIP IF BOTTOM SIGNAL OCCURS SUB# 0 2 SZR JMP OVERH SKIP IF SIGNAL ON MOVL# I I SNC ; GO TO SIGNAL OFF JMP BTOPE ; SET RELAY BIT ON STA 2 @BTRLA FRINT MESSAGE STA Z @BTON ; FUT O IN AC2 SUB 2 2 STA 2 @BTOF JMP SSAMP+4 2~176400 BTOPE: SUB 3 3 STA 3 CBTRLA

; CLEAR RELAY BIT ; FRINT MESSAGE SKIP IF SIGNAL ON ; SET RELAY BIT ; PRINT MESSAGE

1801期

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FILEAR RELAY BIT ; PRINT MESSAGE

3 000044	SSAMB
000012 SSAME:	BLK 10.
000057	TRAP
"000000 TRAP:	Ū
1000017 MASK	17
000001 BTRLA:	GADD RELAY, L
000006 BTON:	GADD ALARMIG
000067 BTOF	GADD ALARM, 7
000000 AURIA	GADD RELAYIO

7~125113 OVERH: MOVL# 1 1 SNC

5-176400 OVOPE: SUB 3 3

STA 2 @BTOF

STA 3 @BTON

JMP SSAMP+4

STA 2 @OVRLA

STA Z ROVEON

STA 2 @OVEOF JMP SSAMP+4

STA 3 COVALA

STA 2 @OVEOF

STA 3 COVEON

JIMP SSAMP+4

JMP OVOPE

SUB 2 2

555AME

1**054016

2*102400

3 040454

+**020452

5~177777

5~020452

1-112414

2~000415

3~125113

1-000406

5.052444

6-052444

7~152400

0~052443

1 000763

311056436

4-052437

5~056435

51000756

011000406

1~052433

21052433

3 152400

+ 052432

5-000747

0 052426

1-056424

2.000742

7-1056425

SSAME

000004 OVEON: 000005 OVEOF: GADD ALARM, 4 GADD ALARM, 5 END

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SWAVE

A/D IJNITS A/D UNITS RATE IN D/A UNITS FATE FATE D/A UNITS RATE NIW. W NI NI SOLVENT SOLVENT VALVE MIN' W NOT SEENOON I CONVERSION Z רברו SOLVENT RATE PLOWRATES INTO ML/MIN DATA FOR FEED VALVE DP CELLS, AND IN. RATE 0E 0 RATE IN RATE LN101 FEED OF SOLVENT LN10d RATE SOLFT, FFLOI, SFLOI, WAIT, TWAIT FEEFT, FHL, FLL, FLG I LNEAR, MTPLY SOLVENT 「このでした」との D/A CHANNEL FOR D/A CHANNEL FOR SOLVENT FEAD , READ 1381 FEED ; SET NORT ЦO DNG STTING 0 SETTING CURRENT **LNBRRUD** FLOWRATES ENG INEER ING READINGS FEED UNIVERSITY 1-129 DECEMBER, 1975 6 INUITED TO READ 上二 REGULAD GFEEAD 1-N000 1-NOOL 2 LIMER FEEFR SWAVE CHENG RE-OB CHANE SFLOI TOJLL LINER LI UM DACV 0000 INEA INEA SOLE VORD DACV NOMAGHER **480** CHENICAL PROGRAM CONVERT に日うし DISPLAY 2 CHANG CHANE SOLFR FEEFR SYGTM SFL01 FFLOI P DELAY EXTD อ MTPLY 0000 NIXI ത്ത TITL 0 0 0 0 0 0 AND NREL 500N 130 ENI **B** 「 大 ENT **801** F divi) LUA ASL. E DA LDA LUA 909 BOD LDA 100 100 100 STA LDA STA ローの L DA LUA POD LOL HO-HOD LDA 384. 209 (D) 0 SOLFT: SOLAD: UNDER FEEFA SOLFR レンセエン READ: SCON: SOLF: TIAW ÷ \$100000.E ~ZE00000, H 2100000.世 141 062023 5 065023 3.020436 母、006444 11、044441 正、024405 S00000. M 202000. 五 0.034431 10.054016 至~020442 3 024442 020440 (7⁻⁰24440 0,062023 1,065023 2'026436 5-044444 6-026436 77/020436 006440 004010.5 292000.00 5,000500 900000 (177777 210900.5 100000.8 125000.41

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1802

SWAVE					
531177540		-160			
000007	FEEAD	GADD DATIJM.	7		
151000056		FCGN			
565 000156	FLON:	110			
57-177543		-157			
5W-177777	LINEAT	LNEAR			
511-000000	SFLOI:	0			
5121000000	FFL01:	0			
		; PROGRAM TO	SHIFT THE F	EED FLONRATE TO A	HIGH
		; LEVEL (FHL)	AND TO A L	OW LEVEL (FLL) ALT	ERNA-
•		; TIVELY.			
		; TWAIT	WAITING	HALF FERIOD IN SE	G.
		; FHL AND FLL	ARE IN D/A	IJN I TS	
		;			
		;			
531034421	FEEFT:	LDA 3 FEEVB-	- L		
54.054016		STA 3 USP			
551020433		LDA O FHL		GET HIGH	LEVEL
516 024431		LDA L FLG		; GET FLAG	VALIJE
571124005		COM I I SNR		;HI-LO?	
760-020431		LDA O FLL		; IF FLAG=0, GET LU	W LEVEL
7111030755		LDA 2 CHANF			
72 072023		DOE 2 DACV		; SET DIA LHANNEL	
731061023		DOA O DACV			
741044423		STA 1 FLG		FAVE FLAD VALUE	
/:31024405 7//0000004	_	LUA I IWALI			
765 0300019	5	LUA Z TIMER			
/// 000024·		CHELY			
		, SYSIM			
011010400	T110 TT.				
M2 000264	IWATI:	IBU.			
W31000762 W44000105		DMP FEEFITZ			
000105	neeur.				
174000002	FEEVE	A BEN IV		284	
201000474	FU0.	278			
244000774		100			
C.C 000274		ENT			
		. 640			

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MOSEX PARAMETER TAPE

WSERS OF THE GOSEX EXECUTIVE MUST CONFIGURE ALL OF THE EXECUTIVE PROGRAMS WITH THIS PARAMETERS TAPE THE USER MAY SPECIFY THE VALUE OF A CERTAIN NUMBER OF THE PARAMETERS HEREIN. THESE PARAMETERS ARE:

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NVARS LPTS TTYS PTPS CASS ATODS PTRS CNTCT RLAYS DTOAS FORT4

UN NO ACCOUNT SHOULD OTHER PARAMETER VALUES BE CHANGED

DISPLACEMENTS IN THE GENERAL WORK SPACE

nwsr	GENRL=	10. ; SIZE OF GENERAL WORK SPACE
NUSR	RETUR=	0 ; RETURN ADDRESS STORAGE
JWSR	RETAD=	4 ; RETURN ADDR STORAGE
JWSR	BYTEP=	2 ; BYTE POINTER STORAGE
IUSR	COUNT=	3 ; COUNTER
JWSR	KOUNT=	4 ; COUNTER
JWSR	WORD1=	5 ; TENPORARY STORAGE
IUSR	WORD2=	6 ; TENPORARY STORAGE
ΨSR	WORDS=	7 ; TEMPORARY STORAGE
UJSR	WORD4=	8. ; TEMPORARY STORAGE
WSR	WORD5=	9. ; TEMPORARY STORAGE

MISPLACEMENTS USED FOR THE LOG AND PLOT ROUTINES

WSR	LOOPI=	GENRL ; LOOP COUNTER
WSR	L00P2=	GENRL+1 ; LOOP COUNTER
WSR	SPOT1=	GENRL+2 ; FOINTER
UJSR	SPOT2=	GENRL+3 ; FOINTER
UISR	NDATA=	GENRL+4 ;# DATA TO LIST
UISR	INTVL=	GENRL+5 ;# SECS BETWEEN RECORDS
UISR	CHANL=	GENRL+6 ; CHANNEL # USED FOR OUTPUT
UISR	LNCNT=	GENRL+7 ; PAGE LINE COUNT
UISR	HOUR=	GENRL+8. ;STARTING HOUR OF RECORD
UISR	MIN=	GENRL+9. ;STARTING MINUTE OF RECORD
UISR	SEC=	GENRL+10. ; STARTING SECOND OF RECORD
UISR	TMSH0=	GENRL+11. COUNTER FOR TIME OF DAY DISPLAY
UISR	MNSIZ=	GENRL+12. ;LOWER RANGE VALUE OF PLOT
UISR	MXSIZ=	GENRL+13. UPPER RANGE VALUE OF PLOT
UISR	DNAME=	GENRL+14. ; BYTE POINTER OF DEVICE FILENAME
UISR	UNIT=	GENRL+15. ; DEVICE UNIT #
UISR	ADSYM=	GENRL+16. ; ADDR OF POINTER TO SYNBOLS LIST
UISR	ADVAL=	GENRL+17. ; ADDR OF POINTER TO VALUES TABLE
UISR	ADBUF=	GENRL+18. ; BYTE POINTER TO WRITTING BUFFER
USR	NVARS=	20. ;# VARIABLES TO BE LOGGED OR PLOTTED
USR .	IDCOD=	O ; OUTPUT TASK DEFINITION CODE
J:SR	TABSA=	1 ; DEVICE STORAGE BLOCK S. A
JISR	IDENT=	2 / TASK IDENTIFICATION NUMBER
JSR	FRIOR=	3 ; TASK PRIORITY

JTPUT TASK DEFINITION CODES

Work		160+321
WSR	LOGFN=	160+331
WSR	L060K=	150+341
WSR	L060S=	160+351
-USR	PLTCL=	180+302
WSR	FLTLF=	180+312
WSR	FLTTT=	160+322
WSR	FLTFN=	180+332
WSR -	FLTDK=	180+342
WSR	PLTCS=	180+352

WNSTANTS WHICH ARE USED TO DEFINE DISPLACEMENTS WR THE DISK MESSAGE ROUTINE

WSR. TMESS= GENRL ; TOTAL NUMBER OF MESSAGES IN DISK FILE WSR NXLGT= GENRL+1 ; MAXINUM MESSAGE LENGTH PERMITTED WSR DSKCL= GENRL+2 ; CHANNEL # TO DISK FILE WSR DSKNM= GENRL+3 ; BYTE POINTER TO DISK FILE NAME WSR DEVCL= GENRL+4 ; CHANNEL # TO OUTPUT DEVICE WSR GENRL+5 ; BYTE POINTER TO OUTPUT DEVICE NAME DEVNM= WSR BUFFR= GENRL+6 ; BYTE POINTER TO BUFFER AREA OF LENGTH MXLGT WSR HIBYF= 0 HIGH ORDER BYTE POINTER TO MESSAGE **WSR** LOBYP= 1 ;LON ORDER BYTE FOINTER TO MESSAGE **WSR** LNGTH= 2 ; MESSAGE LENGTH IN BYTES

(ISPLACEMENTS IN THE USER TASKS DEFINITION BLOCKS. ACH BLOCK IS FOUR WORDS LONG AND EACH USER TASK HAS WE BLOCK DEFINING ITS NECESSARY ATTRIBUTES.

WSR	TSKSA=	0	; S. A. OF THE TASK
WSR	TSKFR=	1	PRIORITY OF THE TASK(1 TO 377-NEVER 0)
WSR	TSKID=	2	; TASK IDENTIFICATION CODE-1 TO 300,
			;NEVER O OR ABOVE 300
WSR	TSKIL=	3	;TASK'S KILL FLAG
	THE TAS	K DEFI	NITION BLOCK IS EXTENDED BY ONE WORD
	IF THE	FORT4	PARAMETER, DEFINED BELOW IS NON-ZERD
WSR	TSKMD=	4	; TASK MODE-IF FORTRAN IV TASK, ZERO
			; - IF ASSEMBLER ONLY, NON-ZERO

HE FOLLOWING CONSTANTS DEFINE THE SYSTEM CONFIGURATION ERO: DEVICE NOT IN THE SYSTEM ON-ZERO: NUMBER OF DEVICES CONFIGURED IN SYSTEM

WSR	LFTS=	1	LINE PRINTER
WSR	TTYS=	1	# TELETYPES ON CPU
WSR -	PTPS=	0	; FAPER TAPE FUNCH
WSR	CASS=	0) CASSETTE FILE
LUSR	ATODS=	10.	; ANALOG TO DIGITAL CONVERTER
USR	PTRS=	0	; FAPER TAPE READER
WSR	CNTCT=	10.	; CONTACT SENSE
USR	RLAYS=	10.	; RELAYS
USR	DTDAS=	10.	; DIGITAL TO ANALOG CONVERTER

WOS DISPLACEMENTS USED BY THE GOSEX PACKAGE

USR	USTP=	12	POINTER TO CURRENT PROGRAM'S UST
USR	TID=	12	; TASK ID IN TASK'S TOB
USR	USTCT=	14	CURRENT TOB POINTER TO PROGRAM'S UST
USR	USTIN=	17	START OF USER NREL CODE IN UST

WRTRAN IV PARAMETER HEN ZERD, ALL TASKS MUST BE IN ASSEMBLER HEN NON-ZERO, THIS PARAMETER SPECIFIES THE MAXIMUM NUMBER OF ARALLEL FORTRAN IV TASKS THAT MAY RUN AT ONE TIME. THE

UUSR UUSR R itus R itus R nuusr Juusr NUSR: iuus R DUUSE NUSE RUISA RUISA RUISA RUISA RUISA RUISA RUISA ESS.R NUSR TUISE. NUISR COINTROLLER 105 P **WSR** WSR WSR **WSR WSR** . MEAN= SPCON= SPA2E= MNICON= MNAZE= OUTHI= DELAY= CASCD= DESTN= AWAIT= SETPT= KP= MNCAT= **大丁I**= SMOTH= LPMES= MNEZA= SFICAPE SPE2A= KT D= SCALE= MESLO= OUTPT= EROR3= HOT TTSUM= BLOCK I= GENRL+7 / INTEGRAL SUM STORAGE GENRL+8 / CURRENT OUTPUT ACTION GENRL+10 / HIGH LIMIT ALARM COUNTER GENRL+10 / HIGH LIMIT ALARM COUNTER GENRL+11 / LOW LIMIT ALARM COUNTER GENRL+12 / FILTER CONSTANT GENRL+12 / FILTER CONSTANT GENRL+13 / ADDR OF RAW DATA GENRL+14 / INPUT LOW LIMIT GENRL+15 / INPUT HIGH LIMIT GENRL+15 / INPUT HIGH LIMIT GENRL+15 / INFUT LOW LIMIT GENRL+15 / INFERVAL BETWEEN SUCCESSIVE ALARMS GENRL+17 / MESSAGE # FOR LOW LIMIT ALARM IN MESSS MG GENRL+18 / INTERVAL BETWEEN SUCCESSIVE ALARMS GENRL+20 / PROPORTIONAL DERIVATIVE GAIN GENRL+21 / FROPORTIONAL DERIVATIVE GAIN GENRL+22 / PROPORTIONAL INTEGRAL GAIN GENRL+23 / SCALE FACTOR WITH ACTION DIRECTION GENRL+25 / CASCADE FLAG = GENRL+26 / ADDR OF DESTINATION DATA BLOCK OR / D/A CHANNEL # = GENRL+28 / LOW LIMIT ON OUTPUT GENRL+28 / LOW LIMIT ON OUTPUT GENRL+34, GENRL+35, GENRL+27. GENRL+28. GENRL+29. GENRL+30. GENRL+37. / MES GENRL+38. / LOO GENRL+36. GENRL+33. GENRL+31. GENAL+6 GENRL+32. GENRL+D GENRL+ GENRE+S GENRL+ GENRIC DISFLACEMENTS 4 ; ENGINEERING 10/A (OR A/D) TO ENGINEERING ; A/D TO ENGINEERING UNITS ; ADDR OF ROUTINE TO CONVERT ; ENGINEERING TO A/D UNITS ; MESSAGE # FOR SETFT CAPTION ; ADDR OF ••• END I ERROR AT -; L00P-ADDR OF CONSTANT BLOCK FOR ; ADDR ERROR SIGN WORK ĥ GAGE Ŷ ΰı SPACEin D-I P-T STORAGE \mathbb{P}_{-1} GOSEX # FOR MEAN ROUTINE TO CONVERT ROUTINE # FOR SETPT CAPTION CONSTANT BLOCK FOR 1 777 2 PARAME WORDS 5 (OR A/D) MESSAGE CONVERT TAPE IN OPCOM NG # IN OPCOM N FROM NON FROM UNITS JUNITS. MEAN FROM FROM IN OFCOM. MG IN OPCOM. NG CONVERSION CONVERSION

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EUSE

FORT4=

. TITE SYMBE

; SYMBL TAPE # 1 OF 2 ; USER BUILT SYMBOL TABLE ; THIS TAPE-GOSEX SYMBOLS

ENT USST, UEST

; EXECUTIVE FUNCTION SYMBOLS

. EXTN START, CEASE, STATS, ENDID, ENDAL, CLEAR, CHNGE, PARAM

. EXTN. TUNER, LOG, PLOT, READY, RELSE, RETAN, CLOCK, PROFL.

; DEVICE SYMBOLS

** ** **

. EXTN CONSL, PRINT, TTYPE, PUNCH, DISK, CASET, READR

. NREL

USER MUST DECLARE THE FOLLOWING AT THE END OF TAPE # 2 SST: LAST SYMBOL-CODE WORD # 1

;			
;	UEST=	EST	
;	USST=	SST	

;ALSO, THE USER MUST INCLUDE THE APPROPRIATE EXTN ;DECLARATIONS FOR TAPE # 2 THIS MAY BE DONE IN TAPE # 2 ITSELF ;IN THIS INSTANCE, IT IS NECESSARY THAT THE EXTERNAL ;DECLARATIONS PRECEED ANY OTHER STATEMENTS

; CODING TABLE FOR RADIX 50 SYMBOLS

	CHARACTER	VALUE(DECIMAL)
	NULL	0
	Ō	1.
	1	2
	2	3
	3	4.
	4	5.
	5	6.
	6	7.
	7	8.
	8	9.
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	A	11.
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	C	13.
	D	14.
	E	15
	F	16.
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	37.	

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FOLLGWING WAY # 2 # 1 Y THE USER IN THE SYMBOL CODE WORD SYMBOL CODE WORD ; SYNEOL Ε ENCODED +001 "AECDE" NAY EE E [D]*40.+[E]*32. [A]*40.+[E]*40.+ SYNEDL

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θ0 THE CHARACTER VALUE OF CODING (#] DECOTES THE COU IN THE TABLE ABOVE I GIVEN , WHERE

WRITTEN OCTOBER, 1974 BY JEAN-PIERRE TREMBLAY DEPARTMENT OF CHEMICAL MCMASTER UNIVERSITY HAMILTON, ONTARID, CANA

ENG INEER ING

CANADA

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SYHBL

	1													5
1	01003 SYMEI											13	35	
	00000,000000	EST:												
5		; EXECUTIVE FU	NCTIONS	LIS	T									2
1	00001 1/////	START	1.0		I SYME	SUL.	VALUE	E-USUALI	LΥ Ε	STABL	ISHED	AT I	LŨĂŨ	
2	00002-107700	36800	-		: SYME		CODE	WURD #	2					
2	00003134773	47611			1 SYPE		うりしと	WUKU #	I					
5	00004 177777													
5	00005-111340	57600												
5	00008 0.3184.3	21411												
í	00010/11/4440	20220												
	00010 11484778	47411												
3	00012-177777	ENDIO	- F										6	
÷	000131061040	25120).											
5	00014*060616	24974	·.											
5	000151177777	ENDAL												
7	000161034700	14784	·.											
6	000171060616	24974	•.											
9	000201177777	CLEAR	t											
D	00021-035200	14976												
1	000221052277	21695	i.											
2	000231177777	CHNGE												
3	000241053340	22240),											
+	000251052050	21544	·.											
2	00026 177777	PARAM	í											
2	00027-034740	14816	•-											
2	00030-122124	42066 TUNED	•											
0	00031 177777	7009A												
0	000331140140	20070 4974A												
i	00034 177777	17204 Lüß												
2	000351000000	0												
3	00036-106571	36217	•											
ş	000371177777	PLOT												
5	000401113000	38400												
6	000411123011	42305												
l	00042177777	READY												
ľ	000431045140	19040	•											
K	00044-130543	45411												
ļ	000451177777	RELSE												
ĥ	00046-111340	37600	•											
6	00047 130556	40422 5575N												
	000514107400													
6	00057 120544	45400		1.										
ĺ	000521177777	40400 CLOCK												
ŀ	00054 041640	17312												
ķ	000551052311	21705												
ł	00056 177777	FROFL	-											
þ	000571051300	21184												
	00060/123371	42745												
ł	2													
I	-	; DEVICES LIST												
į	00061 177777	CONSL												
	00062 111700	37824.												
I	00063-052500	21824.												
I	00064 177777	PRINT												
	00065-075700	31680,												
	00066 123363	427.57												

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000073 00073 00073 00073 00073 00073 00075 00075 00075 00075 00075 00075 00075	00105 00106 00107 00107 00110 00111 001112 001112	00115 00115 00122 00122 00122 00122 00122 00122 00122 00122 00127	80100 80100 80100 80100 80100 80100 80100 80100 80100 80100 80100 80010 800000000	. 19100

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NUTER CONTRACT PARAMETERS PROVIDED TUHLT いたにしたのである C00P SOLFT, FFLOI, SFLOI, WAIT, TU FEEFT, FHL, FLL, FLO FEETC, FNOFP, FTSET, ADFTE SOLTC, SNOFP, STSET, ADSTE SOLTC, SNOFP, STSET, ADSTE FFLOW, OPERP, FFCON, FTEMP I SFLOW, OPERS, SFCON, STEMP APE #2 0F 2 FFL01, SFL01, WATT, FHL, FLL, FL0 101 SOLVENT) CONTROL LOUP FEEFM, CONVE, 1000 CONTROL 1Q 10 +16 101+ 174 N Ю N 0 17 101+ 0+ 10 154 12 JOHLNDO ---114 Ń N N. DUTER 11 IN ----SCHN + ÷ ÷ ŀ + + + + + ÷ *60. *40 *40. \$40 \$40 240 N0 * 01 C4 * 9 N *40 \$ 00 20 \$40 *40 *40. +26, 462 N D * 45.2 \$40° *019 *40 N +40 1254 \$40 154 10.4 10 * (i) * 00 19 1-16. *40. +30. * TEMPERATURE SOLFN, DATUM, GUTBS, NO * \$ **UND** したはいの 10 1 1 1 1 +30. 130. +26. 101+ +14 24 +26 +24 02+ +26. +30. +53 5 20 ID. ंच + 1 4 + 24. + 16. 17. 32 12 TEMPERATURE + ĩst. 9 + ŀ 11. *40. +1 FLOWRATES(FEED A SFLOW 25. *40. +1 29. *40. +1 OPERS + F + ÷ SYMEOU SYMEOU SYMEOU SYTN BUCKTN BUCK SYTN BUCKTN BUCK SYTN BUCKTN BUCK SYTN BUCKTN BU 30. *40. 29. *40. \$N0FP 16. *40. 29. *40. 0TSET 16. *40. 16. *40. FTSET 15. *40. 16. *40. ADFTE 23. *40. *40 ·0+* ×40 *40 30, +40, 440 *40. ×40 今中0 FNOFP 238 *40 238 *40 8700N 231 *40 231 *40 239 *40 239 *40 EXTN FTERF *40 FEETC SOL TO ADT 10. 00 -22 10 0 N こうしく思え ; FEED ... 00155/177777 00156/051500 00157/063731 00160/177777 00161/047300 00162/064315 00163/177777 00171/177777 00172/113640 00173/134476 00173/134476 00175/051500 00177/177777 00177/177777 00200/047300 00201/135015 00202/177777 00201/135015 00202/177777 00204/043415 00205/177777 00205/177777 177777 177777 1113740 1043400 001820. 135726 177777 107640 120137 177777 100000 13640 00154 063147 33715 000000. 04.3416 2 100440 0 134777 177777 2 1 111 100 **EVNB** 4 0 1. 1.59100 -Ó ١. 00153 00210-000210-00213-00214-00215-00216-00216-002216-002216-66 67 00152 N 1D 00200 -Nm 12 0.0223 0:022 00021 2000 1001 1001 1001 1 12. ...

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ビーエビ FEED IQ. 8 00 00+ N. $\overline{0}$ ψ, IC. ++4 オジャ 10 N ιN 15 (1 (N Ge. -12 \mathbf{N} -----------------IN 124 ---N 11 NI + ----I NII ÷ ÷ ÷ ÷ + ÷ ł ÷ ÷ ÷ ÷ ٠ŀ ÷ ÷ t 2 #40. (0) * 40 . 04* N O 04 04 04 69 04 × 32 40 * (1) (1) (1) (1) 440 *02 *40. * 01 7 40 202* 回答 40 40 40 \$40 40 0 ROUT 20.4 ю). 4 * N 114 N. * * * * * * * * (i) * * * * * (i) * (i) * WAVE N -01 * * est. * * ΰÓ. Θ ίų. $\mathbf{\hat{m}}$ +100. ΞÛ. 001 + 100 + + + 10 + 201 + 88 23 51 22 4 21 (h ----16 $\frac{1}{2}$ 1221 水 * 1M * 20 -0 2 0 ----10 2 NO. 151 0+ + 24 N ----N 0+ -而前 + + 100 o + Ŧ +ŀ ÷ F + + + + + ÷ \pm +0+ +Q, + 12 * 440 + 1 255 * 440 + 1 12 * 440 + 1 255 * 440 + 1 255 * 440 + 1 255 * 440 + 1 256 * 440 + 1 259 * 440 + 1 259 * 440 + 1 259 * 440 + 1 250 * 440 + 1 130 * 440 + 1 150 * 440 + 1 160 * 400 + 1 160 * 400 + 100 * 400 * 400 + 100 * 400 * 400 * 400 * 400 * 400 * 400 * 400 * 40 ·ŀ + + ÷ + ÷ Ť *40. 0.1% *40 16 *40. 30LFT 16 *40. 25 *40. 16. *40. 16. *40. 25 *40. 25 *40. 233. *40. 330. *40. 330. *40. 1617 04* 440 *40 SAMPLING MONI 40 40 40. SANF 0. ***** 16. * FLG * Р. 16. 16. ç. 00 (0) ö 10 14 001 1220 00315-00316-

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0007 SYMBL 003171134723 SST: 29.*40.+29.*40.+11. 0000001UEST=EST 0003171USST=SST . END

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FREQUENCY<15 FRUPURITUNAL OFFICATIVE GAIN(+VE)/(40) PROPORTIONAL-DERIVATIVE GAIN(+VE)/(40) PROPORTIONAL-INTEGRAL GAIN(+VE)/(40) SCOLE FACTOR & DIRECTION OF ACTION/(40) E 0 1 2 1 OCURRENT DISPLAY INTERVAL(SECONDS) ORESET DISPLAY INTERVAL(SECONDS): >INTERVAL IMPROPERLY SPECIFIED<15> AVAILABLE DEACTIVAT PLOTTING C01E<40> L0661N5 **JAUK** i SC CURRENT 1/<40> CONFIGURED FOR US DEVICE DEACTIVATED INTERVAL (SECONDS): INTERVAL (SECONDS) /<40) FUNCTION CONFIGURED DEVICES DE < DIRECTORY & FILENAME ETION CONFIRMED ENAME ERROR(15) FILE A CONTROLLER LOOPCIES アリモアリロ INTERVAL (SECONDS) INACTIVE ЧOН Н FUR SYSTEM DEVICESIS> INFUT HIGH ALARM LIMIT-INPUT LOW ALARM LIMIT-CURRENT SET POINT-PROPORTIONAL GAIN(+VE)/<40> +1000. FUNCTION NOT ACTIVE :# FOR FROM SOURCE FILE TASKCIED NOT EQUIPPED TASK CREATED EQUIPPED DVARIABLES LIST-MAX DDEVICE IN USE TO INITIATE E UNIT & FIN T00 L0NG CREATED IN USE 5 INACTIVE 5 ACTIVE NAME: 2>SPECIFIED DEVICE FLUT 21 KILLED EXECUTIVE 2)CHANGE CONFIRMED 2)INTERVAL IMPROPE ALREADY ALREADY DNU LIST(152 DEND OF LIST(152 22ALL CONFIGURED 22DISK DIRECTORY NAME: <12>NDT A CONTROL!
FILTER CONSTANT(0 12>DEVICE IN USE 12>DEVICE NOT EQU 12>DISPLAY MODE: HIGH LIMIT FACTOR: <15><12><40>*?*.<15><12><40> ·<15><12><40> · LOW LINIT Ю Z>DEVICE NOT NOT A USER TASK TASK NUT SEXECUTIVE ビビア 国にとれば READ 2>FILENAME Ш C12>INTERVAL () 1 DTASK IS MEAN-22PL0TTING 0) 14 U) H 7L066ING AN 2>DEVICE TURTUDY **NUNABLE** VCASSET >DEVICE đ LINF AS R CIS>CI2> CI2>NOT A SCALE C12>TASK UTABK 12>USER **Z** TABK Z>USER VTAGK -DON-C CONTROL TONCE NINAN TUATUO TUATLIO TUTFUI iN. N 04 NN 124 N N **IN** NIM 04 \mathbb{N} 154 (1)U 7777 7 v v V. ١Ų 12 シウワワ ∇ :# 0000 0000 0000 0010 20105 9000 30100 0000 0100 2100 0013 0014 0013 00200 0023 0028 0200 1010 1000 1100 0016 0017 0180 61000 0021 0024 0025 0026 0027 032 033 1034 000 9036 8200 6000 0400 1900 2400 0043 00044 0045 00046 30048 0049 00200 00.51 (1)(0)(1) 0031 **1000** 20052 0053 0054 00555 0056 10057 0047 En la

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